

# Optimized Utilization of Kota Stone Slurry Waste in Fly Ash–Based Geopolymer Mortar: A Taguchi-Driven Approach

Pawan Patidar<sup>1,\*</sup>, Pawan Kalla<sup>2</sup>, Pravesh Saini<sup>3</sup>

## Abstract

*The large-scale generation of stone-processing wastes presents a critical sustainability challenge and an opportunity for value-added reuse in construction materials. This study develops a high-performance fly ash geopolymer mortar by partially replacing Class F fly ash with Kota stone slurry waste (KSSW) and optimizing the key mix parameters using a Taguchi design framework. Five governing factors—binder replacement level, NaOH molarity, sodium silicate-to-sodium hydroxide ratio (SS/SH), curing temperature, and alkaline solution-to-binder ratio (A/B)—were systematically optimized using ultrasonic pulse velocity (UPV; larger-the-better) and water absorption (smaller-the-better) as primary compactness and durability indicators, with strength confirmation. The optimum combination (25% KSSW, 12 M NaOH, SS/SH = 1, curing at 60 °C, A/B = 0.5) achieved a UPV of 4380 m/s (+16.7%) and a minimum water absorption of 3.1% (−63.9%) relative to the control, indicating substantial matrix densification. Correspondingly, compressive and flexural strengths reached 25.6 MPa (+143.8%) and 3.95 MPa (+62.6%) at 28 days. Regression modelling and ANOVA confirmed curing temperature as the dominant factor, while excessive activator content adversely affected performance. Durability evaluation of the control and optimum mixes showed progressive degradation under acid exposure, whereas sulphate immersion up to 90 days resulted in net strength gain with early mass increase (pore filling) followed by slight mass loss at longer exposure; drying shrinkage remained comparable between mixes ( $\approx 0.00158$  at 90 days), indicating no shrinkage penalty for the optimized system. Multi-scale characterization (XRD, SEM–EDS elemental mapping, FTIR, XRF, and TGA/DTG–DSC) supported improved gel development and Ca-assisted densification in the KSSW-modified matrix. Overall, the results establish KSSW as a viable supplementary binder and demonstrate that strength-driven statistical optimization enables a resource-efficient, high-performance geopolymer mortar suitable for sustainable construction applications.*

**Keywords:** Geopolymer mortar; Kota stone slurry waste; Taguchi optimization; Ultrasonic pulse velocity; Water absorption.

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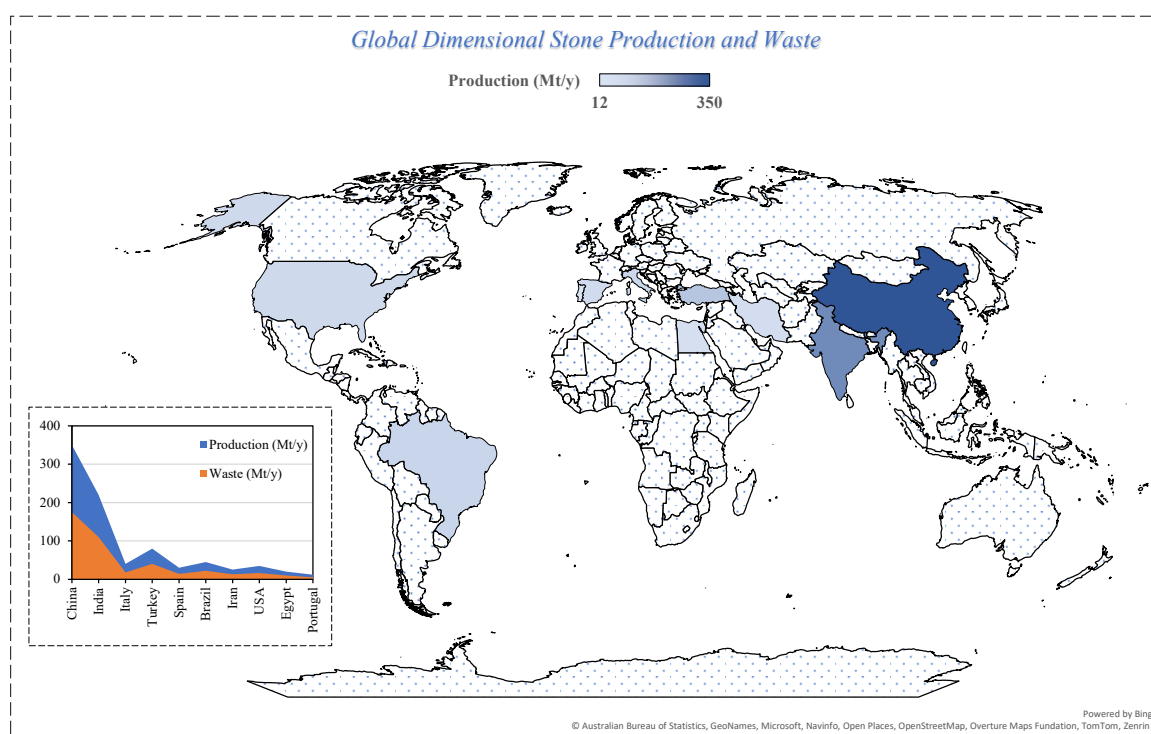
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## INTRODUCTION

The escalating environmental footprint associated with Ordinary Portland Cement (OPC) production has intensified global efforts to develop sustainable, low-carbon construction materials material [1,2]. In this context, geopolymer binders produced through the alkali activation of aluminosilicate precursors have emerged as a promising alternative owing to their significantly lower CO<sub>2</sub> emissions, superior durability, and adaptability across diverse construction applications [3,4]. Among these systems, geopolymer mortars have gained particular

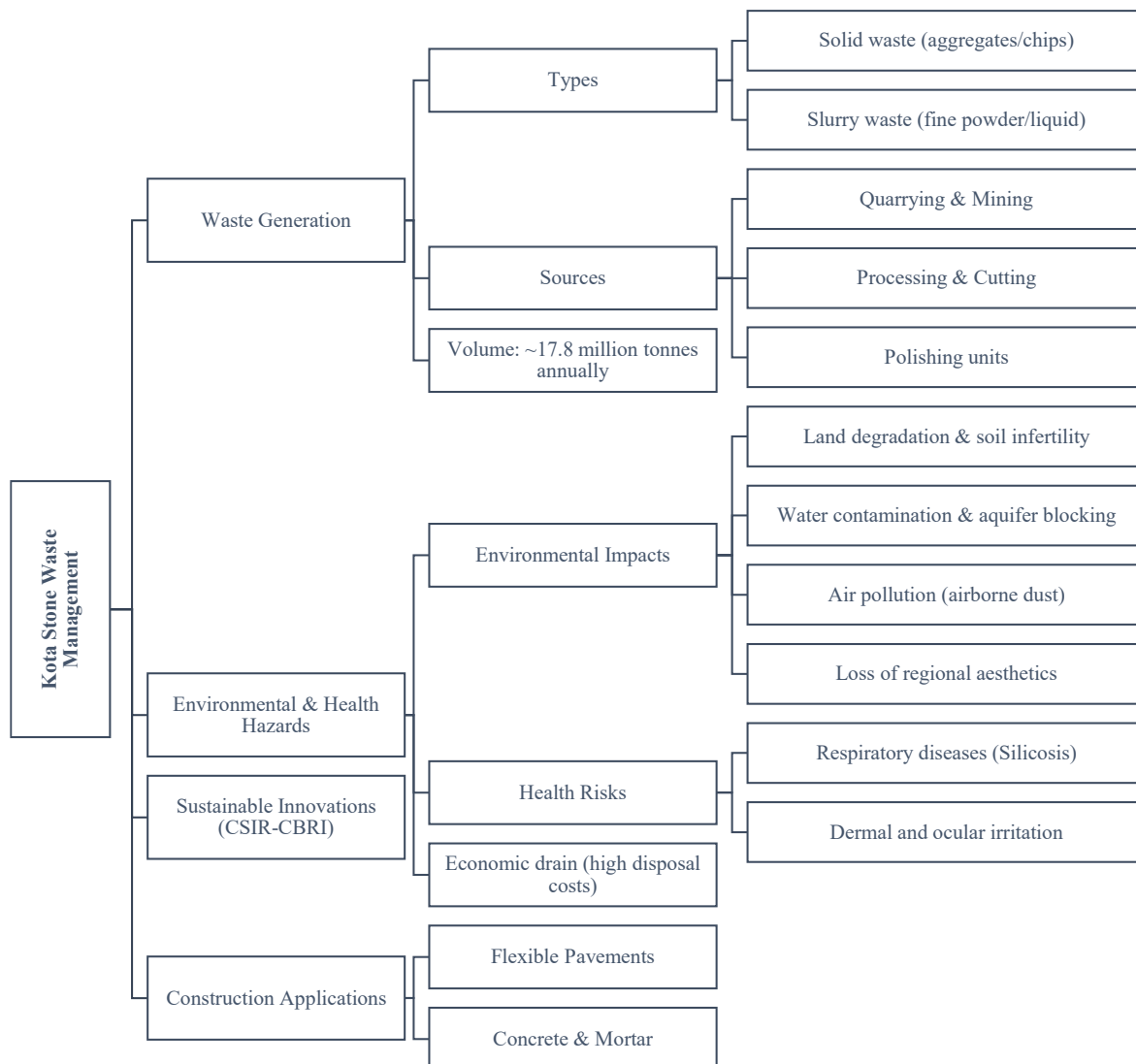
attention for use in masonry, repair works, precast components, and emerging construction technologies [5]. A key driver behind the growing adoption of geopolymer mortars is their ability to incorporate industrial and construction wastes as primary raw materials, thereby reducing landfill disposal while promoting circular-economy practices [4,6]. Fly ash remains one of the most widely investigated geopolymer precursors due to its widespread availability, favourable particle morphology, and high amorphous aluminosilicate content [2]. Numerous studies have demonstrated that fly ash-based geopolymer mortars can achieve high mechanical performance and durability when blended with supplementary materials such as ground-granulated blast furnace slag, rice husk ash, calcined clays, ceramic wastes, and bentonite, which enhance geopolymerization kinetics and microstructural compactness [7–10]. Recent advances further highlight the role of mineral additives and nanomaterials in improving long-term durability and functional performance [11,12].

Beyond precursor selection, the chemistry of the alkaline activator system particularly sodium hydroxide concentration and the sodium silicate-to-sodium hydroxide (SS/SH) ratio critically governs dissolution, polycondensation, gel formation, and pore structure in geopolymer mortars [13–15]. Optimized activator compositions promote the formation of dense N–A–S–H and hybrid C–(N)–A–S–H gels, resulting in enhanced strength and durability, whereas excessive alkalinity may adversely affect workability and microstructural [16]. Curing conditions further influence geopolymer performance; elevated temperatures in the range of 60–90 °C have been shown to accelerate reaction kinetics and strength development in fly ash- and waste-based systems, although excessive heating can induce shrinkage cracking and increased permeability [17–19]. Parallel to these material-level challenges, the global dimensional-stone industry generates enormous quantities of waste during quarrying, cutting, sawing, and polishing operations. Large-scale production of granite, marble, limestone, and related stones for flooring and architectural applications is inherently accompanied by substantial waste streams in the form of quarry rejects, off-cuts, and fine sludge [20]. As illustrated in Figure 1, even conservative waste factors of 0.45–0.50 result in tens to hundreds of million tonnes of stone waste annually, with China (~175 Mt y<sup>-1</sup>) and India (~110 Mt y<sup>-1</sup>) alone accounting for a dominant share of global generation [21]. This waste not only poses serious land, air, and water pollution concerns, but also represents a largely underutilized calcium and silica-rich resource.



**Figure 1.** Global dimensional-stone production and waste generation.

Accordingly, the development of high-volume, performance-oriented pathways to upcycle dimensional stone waste into geopolymer mortars is both an environmental necessity and a circular-economy opportunity. In this context, Kota stone slurry waste (KSSW), a limestone-derived by-product abundantly generated in India, offers significant potential as a supplementary binder in fly ash-based geopolymer mortars. However, systematic optimization of mix design parameters and validation of durability-related performance remain limited. Addressing this gap, the present study integrates Taguchi-based experimental optimization with mechanical, durability, and microstructural evaluation to establish an efficient and sustainable route for valorising KSSW in geopolymer mortar applications. The Kota stone industry generates large volumes of waste in the form of solid chips and fine slurry during quarrying, cutting, and polishing operations.



**Figure 2.** Schematic overview of Kota stone waste generation, sources, environmental and health impacts.

This waste stream, estimated at ~17.8 million tonnes annually, poses serious environmental and health challenges, including land degradation, water contamination, airborne dust, and occupational diseases such as silicosis (See Figure 2). In addition to environmental impacts, improper disposal leads to economic losses due to high handling and landfill costs. Recent sustainability-driven initiatives highlight the need to redirect this waste toward value-added construction applications, particularly in concrete and mortar systems, to mitigate environmental damage while promoting circular economy

practices. Despite extensive research on geopolymer mortars, most studies have focused on conventional aluminosilicate precursors, with limited systematic investigation into the valorization of stone-processing wastes such as Kota stone slurry waste (KSSW) as a supplementary binder. Furthermore, geopolymer mix designs are frequently developed using trial-and-error methods, and the coupled effects of critical parameters including binder replacement level, alkaline activator concentration, alkalinity ratio, curing temperature, and alkaline solution-to-binder ratio remain insufficiently optimized using statistically robust frameworks. To address these gaps, the present study employs a Taguchi-based design of experiments to optimize fly ash–KSSW geopolymer mortar, evaluate its fresh and hardened performance, and elucidate the governing mechanisms through mineralogical characterization.

## MATERIALS AND METHODS

### Fly Ash

Class F fly ash, obtained from the Suratgarh Thermal Power Plant, Rajasthan, India, was used as the primary aluminosilicate precursor in this study. Its suitability for geopolymerization was confirmed through X-ray fluorescence (XRF) analysis (Table 1), which shows a high combined content of  $\text{SiO}_2$  and  $\text{Al}_2\text{O}_3$  ( $\approx 89$  wt.%) and a low  $\text{CaO}$  content ( $\approx 6$  wt.%). This low-calcium composition is characteristic of Class F fly ash and favors alkali activation through dissolution of amorphous aluminosilicate phases, leading predominantly to N–A–S–H gel formation. The fine particle size and predominantly spherical morphology of fly ash further contribute to improved particle packing and a dense geopolymer matrix, which is essential for durability-related properties such as ultrasonic pulse velocity and water absorption.

**Table 1.** Oxide composition of fly ash and Kota stone slurry waste determined by XRF analysis (wt.%).

| Material | $\text{Al}_2\text{O}_3$ | $\text{CaO}$ | $\text{Fe}_2\text{O}_3$ | $\text{MgO}$ | $\text{P}_2\text{O}_5$ | $\text{K}_2\text{O}$ | $\text{SiO}_2$ | $\text{Na}_2\text{O}$ | $\text{TiO}_2$ | LOI  |
|----------|-------------------------|--------------|-------------------------|--------------|------------------------|----------------------|----------------|-----------------------|----------------|------|
| Fly Ash  | 31.56                   | 5.89         | 3.26                    | 1.09         | 0.35                   | 0.87                 | 57.69          | 0.13                  | 1.68           | 0.55 |
| (KSSW)   | 0.87                    | 39.5         | 0.92                    | 1.89         |                        | 0.56                 | 30.59          |                       |                |      |

LOI–Loss on ignition

### Kota Stone Slurry Waste

Kota stone slurry waste (KSSW) was used as a supplementary binder to partially replace fly ash, aiming to valorise stone-processing waste generated in large quantities by cutting and polishing units in Kota, Rajasthan, India. The slurry was collected in wet form, air-dried, oven-dried to remove residual moisture, and ground to obtain a fine powder suitable for geopolymer mortar applications. As summarized in Table 1, XRF analysis confirms that KSSW is predominantly calcium-rich, with high  $\text{CaO}$  content (39.5 wt.%) and moderate  $\text{SiO}_2$  (30.59 wt.%), reflecting its limestone origin. The low alumina content indicates limited primary geopolymer reactivity; however, the fine particle size and carbonate-rich composition enable KSSW to act as an effective micro-filler and nucleation medium. In alkali-activated systems, the calcium contribution supports secondary reaction mechanisms and the formation of hybrid binding phases, contributing to improved matrix densification and early-age performance.

### Natural Aggregates

Natural river sand sourced from the Banas River, Rajasthan, India, was used as the fine aggregate in geopolymer mortar preparation. The sand was clean, free from organic impurities, and brought to a saturated surface-dry (SSD) condition prior to mixing. Sieve analysis conducted in accordance with IS 383 and IS 2386 (Part I) [22,23] confirmed that the sand conformed to Zone III grading, ensuring suitable particle packing and consistency for mortar applications, as illustrated in Figure 3.

### Design of Experiments

A Taguchi design of experiments (DoE) was employed to optimize the geopolymer mortar mixture and to quantify the combined influence of key processing parameters on ultrasonic pulse velocity ( $V_{\text{UPV}}$ )

and water absorption, which are indicators of matrix compactness and durability, respectively. Five control factors were selected based on their established role in geo-polymerization and pore-structure development: binder replacement (B, %), NaOH molarity (M), alkalinity ratio ( $R_{\frac{SS}{SH}}$ ), curing temperature (T), and alkaline solution-to-binder ratio ( $R_{\frac{A}{B}}$ ). The selected factors and their levels are summarized in Table 2. The experimental domain comprised one six-level factor (B) and four three-level factors, resulting in 486 possible combinations ( $6 \times 3^4$ ). To reduce experimental effort while retaining statistical reliability, a mixed-level Taguchi L18 orthogonal array was adopted, which efficiently accommodates one multi-level factor along with multiple three-level factors while preserving orthogonality for main-effect analysis. In this framework, binder replacement denotes the percentage substitution of fly ash by Kota stone slurry waste. Each experimental run was prepared and tested under the specified conditions. Performance evaluation was carried out using the signal-to-noise (S/N) ratio, enabling robust identification of optimal factor levels by minimizing experimental variability.

For UPV, which reflects material compactness and internal continuity, the “larger-the-better” criterion was adopted:

$$\frac{S}{N} = -10 \log_{10} \left( \frac{1}{n} \sum_{i=1}^n \frac{1}{y_i^2} \right) \quad (1)$$

For water absorption, which indicates pore connectivity and durability, the “smaller-the-better” criterion was used:

$$\frac{S}{N} = -10 \log_{10} \left( \frac{1}{n} \sum_{i=1}^n y_i^2 \right) \quad (2)$$

where  $y_i$  is the measured response value (UPV or water absorption) for the  $i$ th experiment, and  $n$  is the number of observations per trial.

**Table 2.** Factors and levels for Taguchi design

| Factors                           |                     | Level1 | Level 2 | Level 3 | Level 4 | Level 5 | Level 6 |
|-----------------------------------|---------------------|--------|---------|---------|---------|---------|---------|
| Binder replacement%               | B (%)               | 5      | 10      | 15      | 20      | 25      | 30      |
| Molarity                          | M                   | 10     | 12      | 14      |         |         |         |
| Alkalinity ratio                  | $R_{\frac{SS}{SH}}$ | 1      | 2       | 3       |         |         |         |
| Temperature                       | T                   | 25     | 60      | 90      |         |         |         |
| Alkaline Solution to Binder Ratio | $R_{\frac{A}{B}}$   | 0.3    | 0.4     | 0.5     |         |         |         |

The selection of control factors and their respective levels in the Taguchi orthogonal array was guided by an extensive review of existing literature, which consistently emphasizes the dominant role of alkaline activator chemistry and curing conditions in governing the performance of fly ash-based geopolymer mortars. Sodium hydroxide (NaOH) molarity is widely recognized as a critical parameter influencing geopolymerization kinetics, pore structure, compactness, and durability-related properties such as ultrasonic pulse velocity (UPV) and water absorption. Previous studies reported that moderate NaOH molarities promote effective dissolution of aluminosilicate species and lead to denser matrices, whereas excessively high molarities can induce rapid precipitation, increased porosity, and reduced performance [24,25]. Accordingly, NaOH molarity levels of 10 M, 12 M, and 14 M were selected to capture the optimum and post-optimum response range [26]. The sodium silicate-to-sodium hydroxide (SS/SH) ratio was incorporated due to its strong influence on silica availability, gel formation, and matrix densification, which directly affect UPV and water absorption. Literature reports indicate that optimized SS/SH ratios enhance the formation of compact aluminosilicate networks, while excessive silica may hinder polymerization and increase pore connectivity [27,28]. Therefore, SS/SH ratios of 1, 2, and 3 were adopted [29]. Curing temperature was selected as a key factor because it strongly controls reaction kinetics, microstructural evolution, and durability-related properties [30]. Elevated curing

temperatures are known to accelerate geopolymerization and enhance matrix compactness, resulting in higher UPV and reduced water absorption; however, excessive temperatures may cause moisture loss and microcracking [31–33]. Thus, curing temperatures of 25 °C, 60 °C, and 90 °C were chosen to represent ambient, moderate, and elevated curing regimes. Finally, the alkaline activator-to-binder (A/B) ratio was included due to its direct influence on pore structure, workability, and permeability characteristics. Prior studies demonstrated that lower A/B ratios Favor matrix densification and reduced water absorption, whereas higher ratios increase pore volume despite improving workability [34,35]. Based on reported optimal ranges, A/B ratios of 0.3, 0.4, and 0.5 were selected [36,37].

### Sample Preparation and Experimental Program

Geopolymer mortar specimens were prepared using Class F fly ash partially replaced with Kota stone slurry waste (KSSW), natural river sand as fine aggregate, and a blended alkaline activator consisting of sodium hydroxide (NaOH) and sodium silicate ( $\text{Na}_2\text{SiO}_3$ ) (See Figure 4). NaOH solutions with molarities ranging from 10–14 M were prepared as per the Taguchi experimental matrix. Dry materials (binder and sand) were first mixed to achieve uniformity, followed by gradual addition of the pre-prepared alkaline activator and mixing water (See Figure 4). Mixing was carried out in a laboratory mixer until a homogeneous and workable mortar was obtained. Fresh mortar was cast into standard moulds and compacted immediately after mixing. After demoulding, specimens were cured under controlled temperatures of 25 °C, 60 °C, and 90 °C. Upon completion of curing, hardened specimens were evaluated for ultrasonic pulse velocity (UPV) to assess internal compactness and continuity, and water absorption to characterize pore structure and durability-related performance. The control mix was prepared using 100% fly ash with a binder-to-sand ratio of 1:3 and equal volumes of NaOH and  $\text{Na}_2\text{SiO}_3$  (1:1).



**Figure 4.** Schematic illustration of material selection, mixing, casting, curing, and testing sequence.

## RESULTS AND DISCUSSION

### Design of Experiment

Based on the Taguchi signal-to-noise (S/N) analysis, the optimum level of each factor was identified from the maximum mean S/N ratio for ultrasonic pulse velocity (UPV) and the minimum mean S/N ratio for water absorption, as summarized in Table 3. For UPV, the binder replacement factor (B) exhibits its highest S/N value at Level 5 (72.23), corresponding to 25% replacement, which was therefore selected as the optimum. Similarly, NaOH molarity (M) attains its maximum S/N at Level 2 (71.94), indicating 12 M as the optimum molarity. The alkalinity ratio ( $R_{SS}^{SH}$ ) shows its highest S/N value at Level 1 (71.87), leading to the selection of  $SS/SH = 1$ . For curing temperature (T), the maximum S/N value occurs at Level 2 (72.20), identifying 60 °C as the optimum curing condition. Finally, the alkaline

solution-to-binder ratio ( $R_{\frac{A}{B}}$ ) achieves its highest S/N at Level 3 (71.96), hence A/B = 0.5 was selected.

For water absorption, which follows the “smaller-the-better” criterion, the minimum mean S/N values in Table 3 confirm the same optimum factor combination. The lowest water absorption corresponds to binder replacement at Level 5 (-16.50), molarity at Level 2 (-16.93), SS/SH at Level 1 (-16.81), curing temperature at Level 2 (-16.48), and A/B at Level 3 (-16.76). The consistency of optimum levels obtained from both UPV and water absorption responses demonstrates that the selected parameter combination simultaneously enhances matrix compactness (higher UPV) and durability (lower water absorption), validating the robustness of the Taguchi optimization. The Taguchi signal-to-noise (S/N) analysis using the “larger-the-better” criterion (Table 3) indicates that UPV is maximized at 25% binder replacement, 12 M NaOH, SS/SH = 1, curing at 60 °C, and A/B = 0.5.

**Table 3.** Response Table for Signal to Noise Ratios.

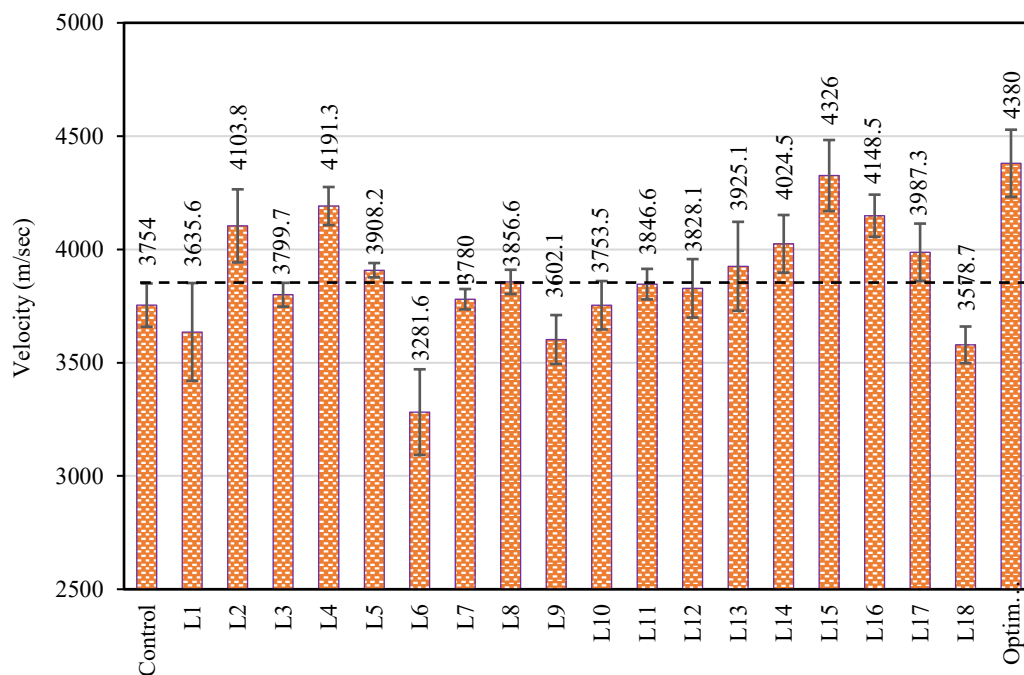
| Factors                   | Level                                  | B (%)  | M      | $R_{\frac{SS}{SH}}$ | T      | $R_{\frac{A}{B}}$ |
|---------------------------|----------------------------------------|--------|--------|---------------------|--------|-------------------|
| Ultrasonic Pulse Velocity | 1                                      | 71.69  | 71.82  | 71.87               | 71.33  | 71.47             |
|                           | 2                                      | 71.54  | 71.94  | 71.71               | 72.20  | 71.75             |
|                           | 3                                      | 71.47  | 71.42  | 71.60               | 71.66  | 71.96             |
|                           | 4                                      | 71.62  |        |                     |        |                   |
|                           | 5                                      | 72.23  |        |                     |        |                   |
|                           | 6                                      | 71.82  |        |                     |        |                   |
|                           | SN <sub>max.</sub> -SN <sub>min.</sub> | 0.76   | 0.52   | 0.27                | 0.87   | 0.49              |
|                           | Rank                                   | 2      | 3      | 5                   | 1      | 4                 |
| Water Absorption          | 1                                      | -17.85 | -17.34 | -16.81              | -17.96 | -18.18            |
|                           | 2                                      | -17.71 | -16.93 | -17.32              | -16.48 | -17.68            |
|                           | 3                                      | -18.50 | -18.35 | -18.49              | -18.18 | -16.76            |
|                           | 4                                      | -17.09 |        |                     |        |                   |
|                           | 5                                      | -16.50 |        |                     |        |                   |
|                           | 6                                      | -17.59 |        |                     |        |                   |
|                           | SN <sub>max.</sub> -SN <sub>min.</sub> | 2.00   | 1.43   | 1.68                | 1.70   | 1.42              |
|                           | Rank                                   | 1      | 4      | 3                   | 2      | 5                 |

The UPV and water-absorption results consistently indicate that the optimized KSSW-based geopolymer mortar achieved a markedly denser and less permeable matrix than the control mix. The control mortar recorded a UPV of 3754 m/s, whereas the optimized mix reached 4380 m/s, representing a ~16.7% increase in pulse velocity, which reflects improved internal continuity and reduced defect content (Figure 5). In parallel, water absorption decreased from 8.6% (control) to 3.1% (optimum), corresponding to a ~64.0% reduction, confirming substantial refinement of pore structure and moisture-ingress resistance (Figure 6). Across the L18 trials, UPV varied widely (3281.6–4326 m/s) while water absorption ranged from 10.5% down to 4.7%, and the optimum mix simultaneously achieved the best durability indicators (highest UPV and lowest absorption), evidencing the effectiveness of the selected factor window (Figure 5 & Figure 6). The inverse UPV–absorption trend is further supported by the fitted relationship ( $R^2 = 0.436$ ), showing that mixes with higher UPV generally exhibit lower water absorption, validating UPV as a rapid, non-destructive proxy for compactness and transport resistance in these geopolymer mortars (Figure 7).

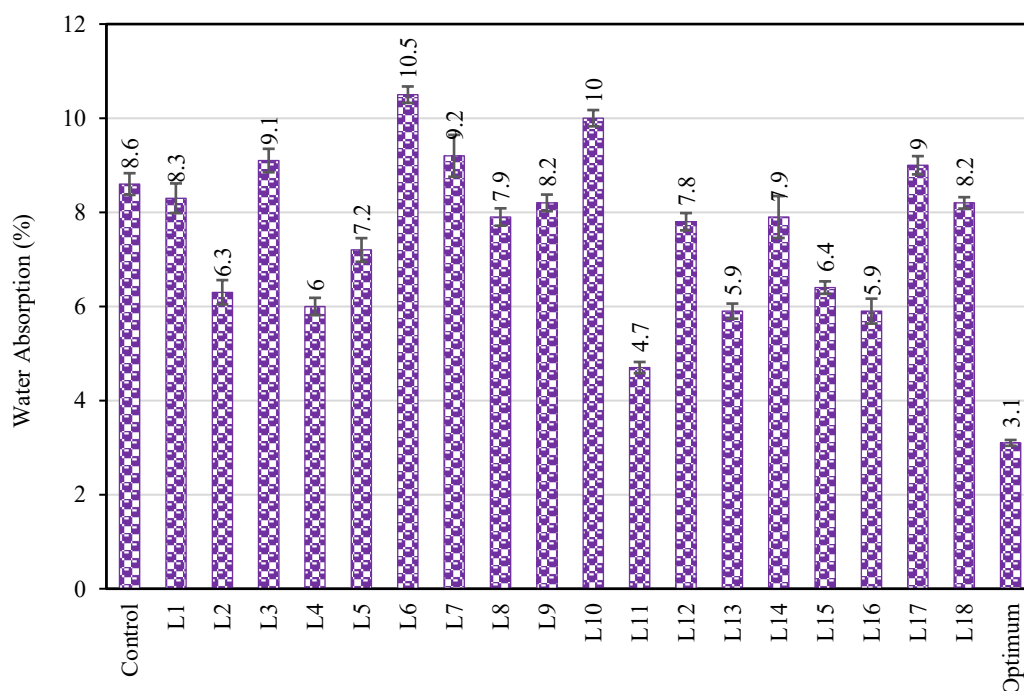
### Enhancement in Mechanical Properties

After identifying the optimum mix through Taguchi optimization, confirmation specimens were cast and directly compared with the control (raw) mix to verify the effectiveness of the selected parameter combination. As illustrated in Figure 8 Improvement in compressive and flexural strength., the

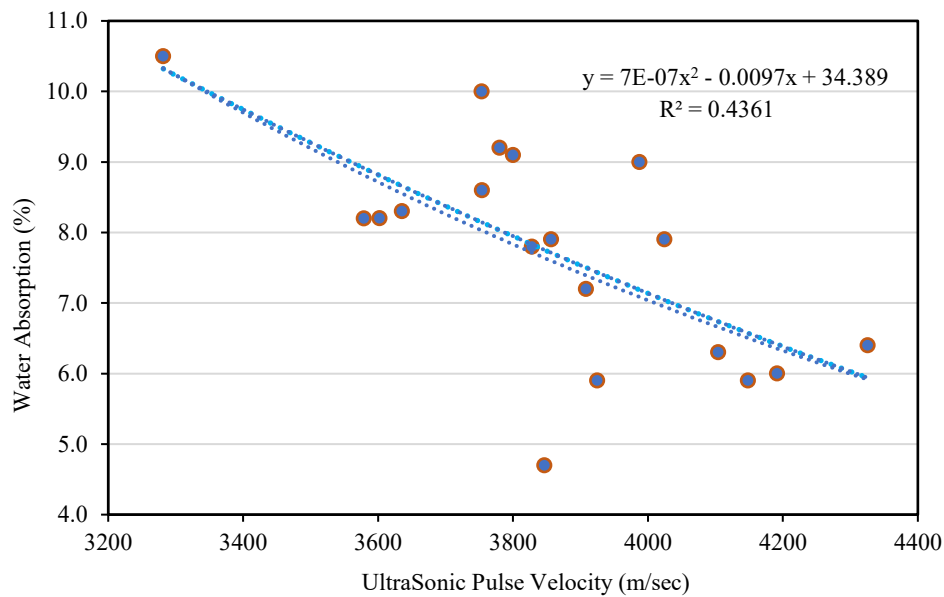
optimized geopolymer mortar exhibits a substantial enhancement in mechanical performance, with compressive strength increasing from 10.5 MPa to 25.6 MPa ( $\approx 143.8\%$  improvement) and flexural strength increasing from 2.43 MPa to 3.95 MPa ( $\approx 62.6\%$  improvement) due to more efficient geopolymerization, improved gel formation, denser microstructure, and stronger binder–aggregate interaction. This pronounced improvement confirms that the optimized combination promotes more efficient geopolymerization, leading to a denser and more homogeneous binder matrix capable of resisting both compressive loads and tensile stresses.



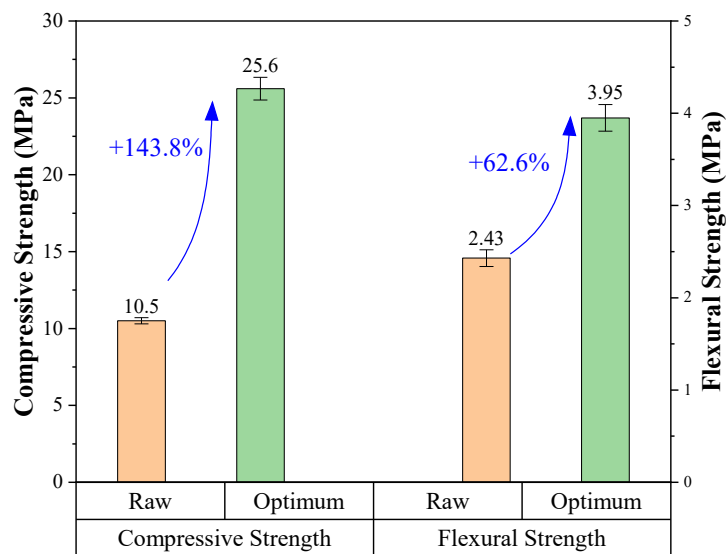
**Figure 5.** Comparison of ultrasonic pulse velocity (m/sec) test results.



**Figure 6.** Comparison of water absorption test results.



**Figure 7.** Relation between ultra sonic pulse velocity and Water Absorption.



**Figure 8.** Improvement in compressive and flexural strength.

### Regression Analysis

Nonlinear regression models were developed to quantify the influence of mix parameters on ultrasonic pulse velocity ( $V_{UPV}$ ) and water absorption ( $W_a$ ) of geopolymer mortar, based on the Taguchi experimental dataset. The proposed equations (Eqs. 3 and 4) incorporate linear, quadratic, and interaction terms of binder replacement ( $B$ ), NaOH molarity ( $M$ ),  $R_{SS}$  ratio, curing temperature ( $T$ ), and alkaline solution-to-binder ratio ( $R_A$ ), enabling robust prediction of durability-related performance indicators. For  $V_{UPV}$ , the regression model demonstrates excellent predictive capability, with an  $R^2$  of 98.93% and adjusted  $R^2$  of 95.45% (Table 4), indicating that the selected variables effectively explain variations in internal compactness. Positive coefficients associated with  $B$ ,  $M$ , and  $T$  confirm their favorable role in enhancing matrix densification and wave propagation velocity, whereas the negative coefficient of  $R_A$  reflects the adverse effect of excess activator on pore continuity. The significant quadratic term  $T^2$  (35.83% contribution,  $p < 0.001$ ) highlights the existence of an optimal curing

temperature beyond which microcracking or moisture loss may reduce UPV. Interaction terms such as  $M \times \frac{R_A}{B}$  and  $M \times T$  were also statistically significant, demonstrating coupled effects between activator concentration and curing conditions.

Ultrasonic Pulse Velocity (m/sec)

$$V_{UPV} = 45 + 50 \times B + 407 \times M - 950 \times \frac{R_{SS}}{SH} + 49.15 \times T - 1353 \times \frac{R_A}{B} - 27.02 \times M^2 + 107.7 \times \frac{R_{SS}^2}{SH} - 0.4858 \times T^2 + 32.80 \times B \times \frac{R_{SS}}{SH} + 1.112 \times B \times T - 435 \times B \times \frac{R_A}{B} - 1.361 \times M \times T + 1002 \times M \times \frac{R_A}{B} \quad (3)$$

Water Absorption (%)

$$W_a = 34.1 - 1.185 \times B - 4.80 \times M - 2.53 \times \frac{R_{SS}}{SH} - 0.2368 \times T + 70.8 \times \frac{R_A}{B} + 0.2170 \times M^2 + 1.107 \times \frac{R_{SS}^2}{SH} + 0.001476 \times T^2 + 0.1001 \times B \times M + 0.00487 \times B \times T - 0.853 \times B \times \frac{R_A}{B} - 4.28 \times M \times \frac{R_A}{B} \quad (4)$$

Similarly, the water absorption model exhibits strong adequacy, with  $R^2 = 95.32\%$  and adjusted  $R^2 = 84.07\%$ , confirming reliable prediction of permeability-related behavior. Negative coefficients for  $B$ ,  $M$ ,  $\frac{R_{SS}}{SH}$ , and  $T$  indicate reduced water uptake with increased binder substitution, optimized alkalinity, and elevated curing temperature, consistent with pore refinement mechanisms. In contrast, the large positive coefficient of  $\frac{R_A}{B}$  confirms that excessive liquid content significantly increases water absorption by promoting capillary porosity. Quadratic and interaction terms particularly  $T^2$  (20.43% contribution) and  $B \times T$  were statistically significant ( $p < 0.05$ ), emphasizing the nonlinear dependency of moisture transport on curing regime and mix composition. The ANOVA results (Table 4) confirm the statistical significance ( $p < 0.05$ ) of both regression models, with low prediction errors (PRESS) and acceptable information criteria (AICc and BIC), validating their suitability for durability-oriented optimization. Overall, the regression and ANOVA analyses demonstrate that UPV and water absorption can be reliably predicted and optimized using nonlinear Taguchi-based models, providing a robust framework for designing dense, low-permeability geopolymer mortars incorporating Kota stone slurry waste.

**Table 4.** Result of ANOVA for compressive strength

|                                   | S                            | R-sq  | R-sq(adj) | PRESS        | R-sq(pred) | AICc   | BIC     | 10-fold S |
|-----------------------------------|------------------------------|-------|-----------|--------------|------------|--------|---------|-----------|
| UPV                               | 53.5561                      | 98.9% | 95.45%    | 447537       | 58.25%     | 437.31 | 210.6   | 144.08    |
| $W_a$                             | 0.629114                     | 95.3% | 84.07%    | 32.8363      | 22.28%     | 179.34 | 51.81   | 1.3241    |
| Ultrasonic Pulse Velocity (m/sec) | Source                       | DF    | Seq SS    | Contribution | Adj SS     | Adj MS | F-Value | P-Value   |
|                                   | Regression                   | 13    | 106054    | 98.93%       | 1060549    | 81581  | 28.44   | 0.003     |
|                                   | B                            | 1     | 66957     | 6.25%        | 44174      | 44174  | 15.40   | 0.017     |
|                                   | M                            | 1     | 86335     | 8.05%        | 14070      | 14070  | 4.91    | 0.091     |
|                                   | $\frac{R_{SS}}{SH}$          | 1     | 43740     | 4.08%        | 58077      | 58077  | 20.25   | 0.011     |
|                                   | T                            | 1     | 71299     | 6.65%        | 179357     | 179357 | 62.53   | 0.001     |
|                                   | $\frac{R_A}{B}$              | 1     | 139651    | 13.03%       | 2123       | 2123   | 0.74    | 0.438     |
|                                   | $M^2$                        | 1     | 71436     | 6.66%        | 38788      | 38788  | 13.52   | 0.021     |
|                                   | $\frac{R_{SS}^2}{SH}$        | 1     | 1173      | 0.11%        | 24926      | 24926  | 8.69    | 0.042     |
|                                   | $T^2$                        | 1     | 384082    | 35.83%       | 507358     | 50735  | 176.8   | 0.000     |
|                                   | $B \times \frac{R_{SS}}{SH}$ | 1     | 3087      | 0.29%        | 76028      | 76028  | 26.51   | 0.007     |
|                                   | $B \times T$                 | 1     | 8011      | 0.75%        | 80271      | 80271  | 27.99   | 0.006     |

|                  | S                          | R-sq | R-sq(adj) | PRESS   | R-sq(pred) | AICc   | BIC   | 10-fold S |
|------------------|----------------------------|------|-----------|---------|------------|--------|-------|-----------|
|                  | $B \times R_{\frac{A}{B}}$ | 1    | 29627     | 2.76%   | 112979     | 11297  | 39.39 | 0.003     |
|                  | M×T                        | 1    | 1765      | 0.16%   | 22455      | 22455  | 7.83  | 0.049     |
|                  | $M \times R_{\frac{A}{B}}$ | 1    | 153385    | 14.31%  | 153385     | 15338  | 53.48 | 0.002     |
|                  | Error                      | 4    | 11473     | 1.07%   | 11473      | 2868   |       |           |
| Water Absorption | Regression                 | 12   | 40.2705   | 95.32%  | 40.271     | 3.3559 | 8.48  | 0.014     |
|                  | B                          | 1    | 1.2652    | 2.99%   | 6.728      | 6.7280 | 17.00 | 0.009     |
|                  | M                          | 1    | 2.0008    | 4.74%   | 1.999      | 1.9985 | 5.05  | 0.075     |
|                  | $R_{\frac{SS}{SH}}$        | 1    | 6.3075    | 14.93%  | 1.043      | 1.0427 | 2.63  | 0.165     |
|                  | T                          | 1    | 0.0018    | 0.00%   | 12.281     | 12.281 | 31.03 | 0.003     |
|                  | $R_{\frac{A}{B}}$          | 1    | 3.9675    | 9.39%   | 6.112      | 6.1123 | 15.44 | 0.011     |
|                  | M <sup>2</sup>             | 1    | 2.5069    | 5.93%   | 2.528      | 2.5283 | 6.39  | 0.053     |
|                  | SS/SH×SS/SH                | 1    | 0.6136    | 1.45%   | 3.248      | 3.2479 | 8.21  | 0.035     |
|                  | T×T                        | 1    | 8.6326    | 20.43%  | 6.289      | 6.2889 | 15.89 | 0.010     |
|                  | B×M                        | 1    | 3.2688    | 7.74%   | 10.517     | 10.516 | 26.57 | 0.004     |
|                  | B×T                        | 1    | 5.7211    | 13.54%  | 5.499      | 5.4991 | 13.89 | 0.014     |
|                  | $B \times R_{\frac{A}{B}}$ | 1    | 2.2944    | 5.43%   | 1.976      | 1.9765 | 4.99  | 0.076     |
|                  | $M \times R_{\frac{A}{B}}$ | 1    | 3.6902    | 8.73%   | 3.690      | 3.6902 | 9.32  | 0.028     |
|                  | Error                      | 5    | 1.9789    | 4.68%   | 1.979      | 0.3958 |       |           |
|                  | Total                      | 17   | 42.2494   | 100.00% |            |        |       |           |

### Validation of Regression Equation

A compares the experimentally measured and regression-predicted values of ultrasonic pulse velocity ( $V_{UPV}$ ) and water absorption ( $W_a$ ) for all 18 Taguchi trial mixes (Table 5). The results demonstrate a close agreement between experimental and predicted responses, confirming the robustness and predictive capability of the developed nonlinear regression models. For UPV, the percentage deviation between measured and predicted values remains low, generally within  $\pm 5\%$ , with several mixes exhibiting errors below 2%. This indicates that the regression model effectively captures the combined effects of binder replacement, activator chemistry, and curing temperature on matrix densification and wave propagation characteristics. Similarly, the water absorption model shows good predictive accuracy, with most deviations below 6% and only a few mixes approaching  $\sim 10\%$ , which is acceptable for durability-related properties. The slightly higher scatter in WA predictions compared to UPV is attributed to the greater sensitivity of water absorption to pore connectivity and microstructural heterogeneity. Overall, the low prediction errors across both responses confirm that the developed regression equations reliably describe the influence of mix parameters on internal quality and durability indicators of geopolymers, making them suitable for performance prediction and mix optimization within the investigated domain (Table 5).

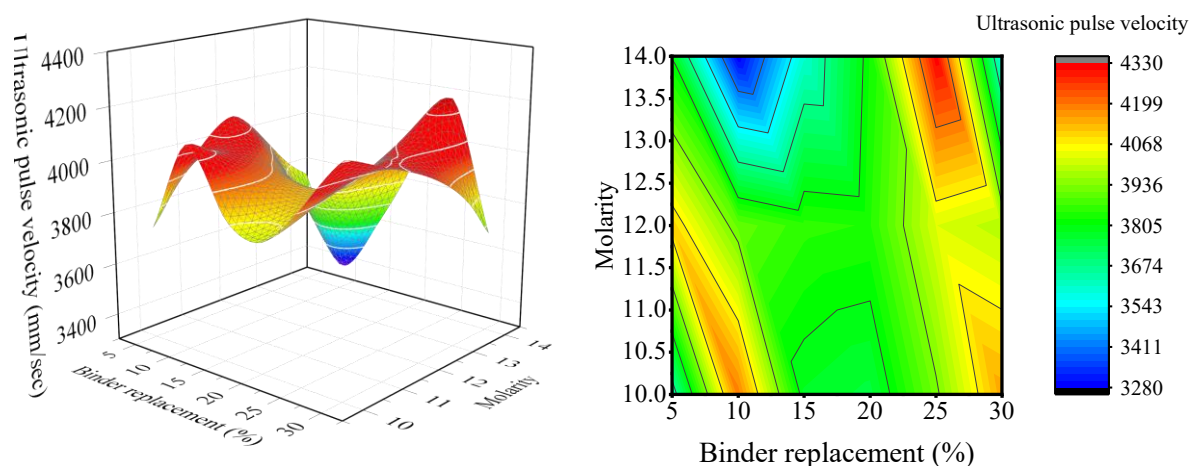
**Table 5:** Comparison between experimental and regression-predicted compressive and flexural strength.

| Sample Cast | B% | M  | $R_{\frac{SS}{SH}}$ | Temp | $R_{\frac{A}{B}}$ | $V_{UPV}$ act | $V_{UPV}$ Pred | $\Delta v$ | $W_a$ act | $W_a$ Pred | $\Delta w$ |
|-------------|----|----|---------------------|------|-------------------|---------------|----------------|------------|-----------|------------|------------|
| L-1         | 5  | 10 | 1                   | 25   | 0.3               | 3635.6        | 3656.18        | 0.24%      | 8.3       | 8.19       | 1.36%      |
| L-2         | 5  | 12 | 2                   | 60   | 0.4               | 4103.79       | 4099.12        | 1.94%      | 6.3       | 5.83       | 8.00%      |
| L-3         | 5  | 14 | 3                   | 90   | 0.5               | 3799.63       | 3832.44        | 2.81%      | 9.1       | 9.03       | 0.78%      |
| L-4         | 10 | 10 | 1                   | 60   | 0.4               | 4191.27       | 4176.22        | 1.31%      | 6         | 6.35       | 5.55%      |
| L-5         | 10 | 12 | 2                   | 90   | 0.5               | 3908.13       | 3904.86        | 5.00%      | 7.2       | 7.76       | 7.21%      |
| L-6         | 10 | 14 | 3                   | 25   | 0.3               | 3281.59       | 3274.66        | 2.15%      | 10.5      | 10.89      | 3.62%      |

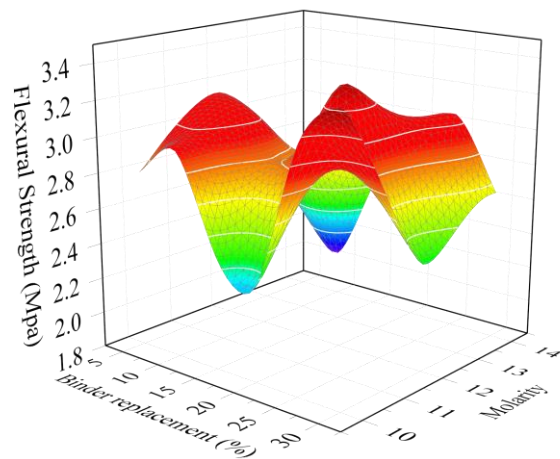
| Sample Cast | B% | M  | $R_{SS}^{SH}$ | Temp | $R_{\frac{A}{B}}$ | $V_{UPV}^{act}$ | $V_{UPV}^{Pred}$ | $\Delta v$ | $W_a^{act}$ | $W_a^{Pred}$ | $\Delta w$ |
|-------------|----|----|---------------|------|-------------------|-----------------|------------------|------------|-------------|--------------|------------|
| L-7         | 15 | 10 | 2             | 25   | 0.5               | 3779.94         | 3750.68          | 5.50%      | 9.2         | 8.84         | 4.08%      |
| L-8         | 15 | 12 | 3             | 60   | 0.3               | 3856.54         | 3848.22          | 7.28%      | 7.9         | 7.85         | 0.69%      |
| L-9         | 15 | 14 | 1             | 90   | 0.4               | 3602.03         | 3581.64          | 0.03%      | 8.2         | 7.71         | 6.40%      |
| L-10        | 20 | 10 | 3             | 90   | 0.4               | 3753.5          | 3752.32          | 4.01%      | 10          | 10.28        | 2.71%      |
| L-11        | 20 | 12 | 1             | 25   | 0.5               | 3846.54         | 3910.15          | 3.16%      | 4.7         | 5.28         | 10.93%     |
| L-12        | 20 | 14 | 2             | 60   | 0.3               | 3828.1          | 3873.66          | 5.21%      | 7.8         | 8.22         | 5.15%      |
| L-13        | 25 | 10 | 2             | 90   | 0.3               | 3925.06         | 3937.02          | 3.36%      | 5.9         | 6.17         | 4.40%      |
| L-14        | 25 | 12 | 3             | 25   | 0.4               | 4024.47         | 3997.65          | 1.58%      | 7.9         | 7.82         | 1.05%      |
| L-15        | 25 | 14 | 1             | 60   | 0.5               | 4325.98         | 4299.66          | 3.19%      | 6.4         | 6.61         | 3.13%      |
| L-16        | 30 | 10 | 3             | 60   | 0.5               | 4148.45         | 4177.92          | 1.07%      | 5.9         | 5.73         | 2.97%      |
| L-17        | 30 | 12 | 1             | 90   | 0.3               | 3987.25         | 3987.16          | 4.97%      | 9           | 8.76         | 2.76%      |
| L-18        | 30 | 14 | 2             | 25   | 0.4               | 3578.64         | 3578.66          | 0.23%      | 8.2         | 8.06         | 1.70%      |

### Interaction Effect of Binder Replacement and NaOH Molarity

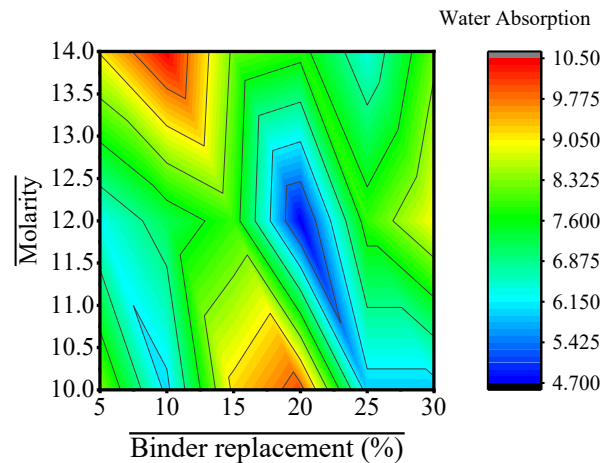
The combined effect of binder replacement (%) and NaOH molarity on the internal quality and durability of geopolymer mortar is illustrated through the 3D surface and contour plots for ultrasonic pulse velocity (UPV) and water absorption (Figure 9, Figure 10, Figure 11, & Figure 12). For UPV, an increase in molarity from 10 M to an intermediate range ( $\approx 12$ – $13$  M) results in a marked rise in pulse velocity across all binder replacement levels, indicating enhanced matrix densification due to improved dissolution of aluminosilicate species and formation of a compact geopolymeric gel network. However, at higher molarity (14 M), UPV decreases for several binder contents, suggesting microstructural discontinuities caused by excessive alkalinity and rapid reaction kinetics. The highest UPV values are consistently observed at intermediate binder replacement levels ( $\approx 20$ – $25\%$ ) combined with moderate molarity, defining an optimum densification zone. A complementary trend is observed for water absorption, where absorption decreases with increasing molarity up to an optimum level and then increases at higher molarity, reflecting increased pore connectivity at excessive alkali concentrations (Figure 11, & Figure 12). Minimum water absorption occurs within the same binder–molarity window that yields maximum UPV, confirming that mixes with higher pulse velocity possess a denser and less permeable microstructure. The inverse relationship between UPV and water absorption is therefore reinforced by the interaction plots, demonstrating that optimal KSSW incorporation coupled with controlled alkalinity improves both internal compactness and durability performance of geopolymer mortar.



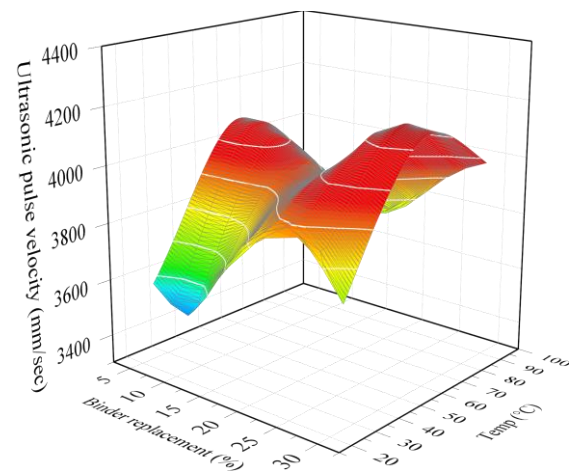
**Figure 9.** 3D surface plot showing the interaction effect of binder content and molarity effect of binder content and molarity for ultrasonic pulse velocity.



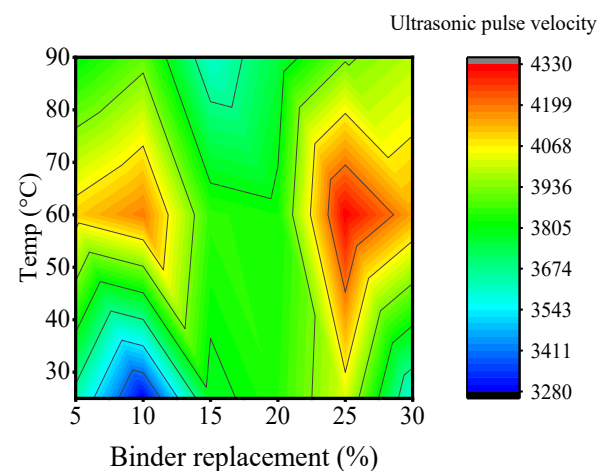
**Figure 11.** 3D surface plot showing the interaction effect of binder content and molarity for water absorption.



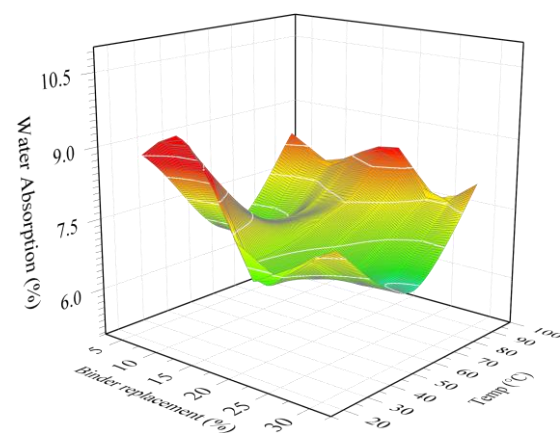
**Figure 12.** 2D contour plot showing the interaction effect of binder content and molarity for water absorption.



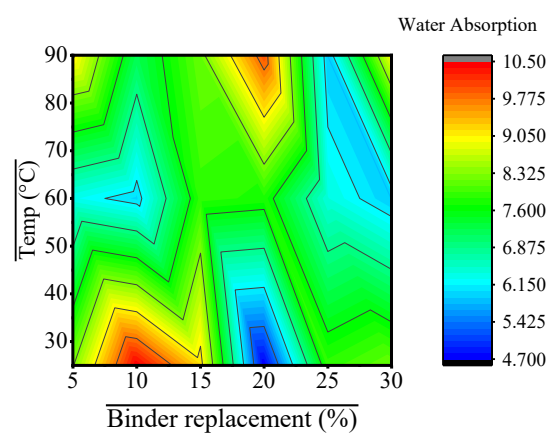
**Figure 13.** 3D surface plot showing the interaction effect of binder content and temperature for ultrasonic pulse velocity.



**Figure 14.** 2D contour plot showing the interaction effect of binder content and temperature for ultrasonic pulse velocity.



**Figure 15.** 3D surface plot showing the interaction effect of binder content and temperature for water absorption.



**Figure 16.** 2D contour plot showing the interaction effect of binder content and temperature for water absorption.

### Interaction Effect of Binder Replacement (%) and Curing Temperature

The coupled influence of binder replacement (%) and curing temperature on ultrasonic pulse velocity (UPV) and water absorption is illustrated by the 3D response surfaces and contour plots (Figure 13, Figure 14, Figure 15, & Figure 16). For UPV, increasing the curing temperature from ambient to moderate levels significantly enhances pulse velocity across most binder replacement ranges, reflecting improved geopolymerization kinetics, accelerated gel formation, and increased matrix compactness. However, at higher temperatures ( $\approx 90^\circ\text{C}$ ), UPV gains diminish or decline for some mixes, particularly at low or excessive KSSW contents, likely due to rapid moisture loss and microstructural discontinuities. The contour plots indicate a clear optimum UPV zone at moderate curing temperature ( $\sim 60^\circ\text{C}$ ) combined with intermediate binder replacement ( $\approx 20\text{--}25\%$ ), where pulse velocity reaches the highest bands, signifying a dense and homogeneous microstructure. A complementary trend is observed for water absorption. Moderate thermal curing markedly reduces water absorption by promoting pore refinement and improved gel continuity, whereas excessive curing temperatures lead to localized increases in absorption, attributed to thermal shrinkage and microcrack formation. The lowest water absorption values are consistently concentrated within the same  $20\text{--}25\%$  binder replacement and  $\sim 60^\circ\text{C}$  curing window, confirming enhanced durability performance under these conditions. Overall, the interaction plots demonstrate that curing temperature and binder replacement act synergistically, and optimal durability and internal quality are achieved only when thermal activation is balanced with appropriate KSSW incorporation, avoiding under- or over-activation of the geopolymer matrix.

### Comparison with previous studies

The mechanical performance of the optimized KSSW–fly ash geopolymer mortar with stone- and limestone-based waste mortars reported in the literature and with relevant codal provisions (Table 6). The optimized mix developed in this study (25% KSSW replacement) achieved 25.6 MPa compressive strength and 3.95 MPa flexural strength at 28 days, demonstrating performance comparable to or higher than several geopolymer and cement mortars incorporating dimensional-stone wastes. Compared with limestone-modified geopolymer systems, the present mortar significantly outperforms geopolymer mortar incorporating limestone powder with FA+BFS (16.16 MPa compressive and 2.88 MPa flexural strength), highlighting the effectiveness of KSSW combined with optimized alkaline activation and curing (Table 6) [38]. While higher strengths have been reported for calcium-rich geopolymer systems such as one-part lime-activated construction-and-demolition waste mortars and calcined dolomite powder geopolymer mortars ( $\approx 42$  MPa compressive strength), these systems employ different precursor chemistries and activation mechanisms that inherently promote higher early strength [39,40]. Relative to cement-based mortars incorporating stone wastes, the optimized geopolymer mortar exhibits superior strength compared with ornamental stone processing waste cement mortar and comparable performance to stone dust–waste marble aggregate mortars, while remaining below high-strength marble powder–fly ash cement mortars [41–43].

**Table 6.** Comparison of present study with previous literature and international standards.

| Source                             | Waste material                           | Binder system                                                                      | Replacement level                     | Curing age | Compressive strength (MPa) | Flexural strength (MPa) |
|------------------------------------|------------------------------------------|------------------------------------------------------------------------------------|---------------------------------------|------------|----------------------------|-------------------------|
| This Study                         | Kota Stone Slurry Waste (KSSW)           | Fly ash–KSSW geopolymer mortar (NaOH + Na <sub>2</sub> SiO <sub>3</sub> activator) | 25% KSSW                              | 28 days    | 25.6                       | 3.95                    |
| (Aslan and Erkan 2024) [38]        | Limestone powder (LP) (with FA + BFS)    | Geopolymer mortar (FA + BFS + LP)                                                  | Mix No. 9: LP = 21.66% (binder blend) | 28 days    | 16.16                      | 2.88                    |
| (Moreira <i>et al.</i> 2022a) [41] | Ornamental stone processing waste (OSPW) | Cement mortar (CP V-ARI)                                                           | 30% OSPW                              | 28 days    | 12.56                      | 3.12                    |

|                                   |                                                              |                                           |                                 |                |              |     |
|-----------------------------------|--------------------------------------------------------------|-------------------------------------------|---------------------------------|----------------|--------------|-----|
| (Woźniak <i>et al.</i> 2022) [42] | Stone dust + waste marble aggregate                          | Cement                                    | 10%                             | 28 days        | 20.54        | 4.5 |
| (Kogbara <i>et al.</i> 2024) [39] | Construction & demolition waste (CDW) (mineral waste stream) | One-part lime-activated geopolymer mortar | 55%                             | 28 days        | 42.0         | 5.8 |
| (Youssif <i>et al.</i> 2025) [44] | Brick-and-stone cutting waste (BSCW)                         | Ambient-cured geopolymer mortar           | 25%                             | Ambient-cured; | 32.76        | -   |
| (Shaaban <i>et al.</i> 2025) [40] | Calcined dolomite powder (CDP)                               | Geopolymer mortars                        | 45%                             | 28 Days        | 42.35        | 4.3 |
| (Doğan and Demir 2021) [43]       | Marble powder + fly ash (cement mortar)                      | Cement mortar (CEM I 42.5R)               | 20%                             | 28 Days        | 49.5         | 9.5 |
| (IS 2250 1981) [45]               |                                                              | Cement:sand mortar                        | 1:3 corresponds to grade MM 7.5 | 28 days        | $\geq 7.5$   |     |
| (ASTM C270 2025) [46]             |                                                              | Masonry mortar classification             |                                 | 28-day         | Type M: 17.2 |     |
|                                   |                                                              |                                           |                                 |                | Type S: 12.4 |     |
|                                   |                                                              |                                           |                                 |                | Type N: 5.2  |     |
|                                   |                                                              |                                           |                                 |                | Type O: 2.4  |     |

Importantly, from a codal perspective, both the control and optimized geopolymer mortars exceed the IS 2250 minimum strength requirement ( $\geq 7.5$  MPa), and the optimized mix comfortably surpasses the compressive strength thresholds defined for masonry mortars under ASTM C270, confirming its suitability for structural masonry and repair applications [45,46].

## CONCLUSIONS

This study optimized a Kota stone slurry waste (KSSW)–modified fly ash geopolymer mortar using a Taguchi-based experimental framework and validated performance through regression modelling, ANOVA, and non-destructive durability indicators. The principal conclusions are summarized as follows:

- The optimal geopolymer mortar composition was achieved at 25% KSSW replacement, 12 M NaOH, SS/SH = 1, curing at 60 °C, and A/B = 0.5, consistently identified as optimal across UPV and water absorption.
- The UPV increased from 3754 m/s (control) to 4380 m/s (optimum), corresponding to an improvement of approximately +16.7%, indicating significantly enhanced matrix densification.
- Water absorption decreased from 8.6% (control) to 3.1% (optimum), representing a substantial ~64% reduction, confirming refinement of pore structure and improved resistance to moisture ingress.
- The optimized mix achieved 25.6 MPa compressive strength and 3.95 MPa flexural strength at 28 days, reflecting increases of +143.8% in compression and +62.6% in flexure relative to the control mortar.
- Regression models incorporating linear, quadratic, and interaction terms accurately predicted UPV and water absorption, with ANOVA identifying curing temperature (T), NaOH molarity (M), and activator-to-binder ratio (A/B) as dominant contributors. Prediction errors remained within acceptable limits (<10%), confirming model robustness.
- Response surface analysis demonstrated that moderate alkalinity ( $\approx 12$  M) combined with intermediate KSSW replacement ( $\approx 20$ –25%) and controlled thermal curing ( $\approx 60$  °C) maximized UPV while minimizing water absorption, emphasizing the importance of balanced reaction kinetics and microstructural stability.

Overall, the findings demonstrate that KSSW can be effectively valorized as a supplementary binder in geopolymer mortar, enabling substantial improvements in strength, durability, and internal quality. The integration of statistical optimization with durability-oriented indicators provides a robust, resource-efficient pathway for sustainable geopolymer mortar design. Future research should focus on long-term durability, environmental exposure resistance, and life-cycle CO<sub>2</sub> assessment to support field-scale adoption and standardization.

## AUTHOR CONTRIBUTIONS

All authors contributed to the development of manuscript.

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## Notations

The following symbols are used in this paper:

- $R_{\frac{A}{B}}$  – Alkaline solution-to-binder ratio
- Adj MS – Adjusted mean square (ANOVA)
- Adj SS – Adjusted sum of squares (ANOVA)
- AI – Artificial intelligence (only if used; otherwise delete)
- AICc – Corrected Akaike information criterion
- Al<sub>2</sub>O<sub>3</sub> – Aluminium oxide
- ANOVA – Analysis of variance
- ASTM – ASTM International (formerly American Society for Testing and Materials)
- B% – Binder replacement / binder replacement percentage
- BFS – Blast furnace slag (as used in literature comparison table)
- BIC – Bayesian information criterion
- BSCW – Brick-and-stone cutting waste
- BR – Binder replacement (remove if you already use B or B%)
- CaCO<sub>3</sub> – Calcium carbonate (calcite)
- CaMg(CO<sub>3</sub>)<sub>2</sub> – Dolomite
- CaO – Calcium oxide
- C–A–S–H – Calcium–alumino–silicate–hydrate
- C–S–H – Calcium–silicate–hydrate
- CI – Crystallinity index
- CO<sub>2</sub> – Carbon dioxide
- CP V-ARI – Portland cement type CP V-ARI (Brazilian designation, as in cited literature)
- CPWD – Central Public Works Department (India)
- Cu–K $\alpha$  – Copper K-alpha radiation
- CDP – Calcined dolomite powder
- CDW – Construction and demolition waste
- DF – Degrees of freedom
- DOE – Design of experiments
- FA – Fly ash
- GGBFS – Ground granulated blast furnace slag
- GP – Geopolymer
- IS – Indian Standard
- KBr – Potassium bromide
- KSSW – Kota stone slurry waste
- LP – Limestone powder

- M – Molarity (NaOH concentration, mol/L)
- MS – Mean square (ANOVA)
- Na<sub>2</sub>SiO<sub>3</sub> – Sodium silicate
- NaOH – Sodium hydroxide
- N–A–S–H – Sodium–alumino–silicate–hydrate gel
- C-(N)-A–S–H – Calcium-(sodium)-alumino–silicate–hydrate (hybrid gel)
- OPC – Ordinary Portland cement
- OSPW – Ornamental stone processing waste
- Q – Quartz
- R<sup>2</sup> – Coefficient of determination
- R<sup>2</sup>(adj) – Adjusted coefficient of determination
- R<sup>2</sup>(pred) – Predicted coefficient of determination
- S – Standard deviation of regression (model “S”)
- Seq SS – Sequential sum of squares (ANOVA)
- SH – Sodium hydroxide solution
- S/N – Signal-to-noise ratio (Taguchi)
- SS – Sodium silicate solution
- $R_{SH}^{SS}$  – Sodium silicate to sodium hydroxide ratio
- T – Curing temperature
- 2θ – Diffraction angle

#### DATA AVAILABILITY STATEMENT

Data will be available on request.

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