

# From Molecular Mechanics to Nanocomposites: Engineering Polytetrafluoroethylene (PTFE) for High-performance Coating Applications

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## Abstract

*Polytetrafluoroethylene (PTFE) exhibits remarkable chemical inertness, hydrophobicity, antifriction, self-lubrication, and high-temperature resilience, making it an ideal polymer for various industrial applications, including coatings for medical implants, machinery parts, and corrosion-resistant structures. This study presents a comprehensive analysis of PTFE's structural properties, focusing on helix reversals within its helical carbon-fluorine chains and their effects on mechanical behavior. The phase transitions of PTFE at temperatures from ambient to high (198°C and above) were analyzed, revealing significant insights into molecular conformations and their impact on thermal and mechanical stability. The performance of PTFE-based nanocomposites, such as TiO<sub>2</sub>-PTFE, was evaluated, demonstrating enhanced antibacterial and anticorrosive properties on stainless steel substrates. By comparing plasma-based coating techniques on carbon steel, this work emphasizes the unique tribological advantages of PTFE in reducing wear and friction in engineering applications. Future developments in PTFE applications in the technical and medicinal domains are made possible by these results.*

**Keywords:** Antibacterial properties, helix reversals, nanocomposites, phase transitions, tribology

## INTRODUCTION

Polytetrafluoroethylene (PTFE), also known commercially as Teflon™, has established itself as a unique polymer with extensive applications due to its exceptional chemical stability, low friction, and high resistance to heat. Structurally, PTFE is composed of long-chain carbon-fluorine bonds arranged in a helical formation, which contributes to its inherent non-reactivity and hydrophobicity. The molecular dynamics of PTFE, specifically helix reversals and phase transitions, play a crucial role in its material properties, making it suitable for applications ranging from engineering components to medical implants. Recent advancements have focused on enhancing PTFE's applications in coatings, leveraging its low friction and self-lubrication characteristics. Notably, TiO<sub>2</sub>-PTFE nanocomposites have emerged as promising candidates for antibacterial and anticorrosion coatings on surgical-grade stainless steel, extending PTFE's utility to biomedical applications.

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Furthermore, the tribological properties of PTFE, including its ability to function effectively under high-pressure and low-friction conditions, make it a versatile choice for machinery parts, such as bearings, shafts, and gears. Studies involving PTFE coatings applied via plasma spraying on carbon steel substrates have demonstrated substantial

reductions in wear and an increase in surface durability. Despite these advantages, PTFE's inherent limitations, such as its susceptibility to creep and restricted weldability, present challenges in high-stress applications. This study examines the physicochemical and tribological properties of PTFE, particularly in the context of TiO<sub>2</sub>-PTFE nanocomposites, molecular dynamics of helix reversals, and phase behavior under varied temperatures. We also explore the potential of PTFE coatings for enhancing joint clearances and wear resistance in spherical joint systems, addressing a vital need in mechanical engineering.

## PTFE

In terms of high chemical inertness, good hydrophobicity, antifriction properties, self-lubrication properties, and high-temperature resistance, polytetrafluoroethylene (PTFE) is a class of polymers. PTFE has excellent non-stick properties, a relatively high melting temperature of 325°C, and is chemically inert. The rich phase behavior of polytetrafluoroethylene (PTFE), which is made up of an organic helical chain structure, was created from a chemically simple polymer. Differential long-chain carbon-fluorine molecule orientations have been investigated under various conditions. Chain originated crystal morphology differences have been defined by X-ray studies on high temperature treated PTFE. The Du Pont Company used the trade name Teflon™ for PTFE. Below 300°C, PTFE is an inert, stable, and non-hazardous substance. The low friction of PTFE has made it suitable for use in appliances acting beneath sliding parts, as well as in plain bearings, gears, slide plates, seals, gaskets, and bushings. As an engineering plastic, PTFE has low friction as well as chemical, high temperature, and weathering resistance. Tape, fabric, and tubing made of PTFE are available. Antibacterial TiO<sub>2</sub>-PTFE nanocomposite coatings have prevented corrosion and bacterial contamination of 316 L stainless steel implants. The functionalized stainless-steel substrate with catechol groups has been coated with the descriptive issued scheme, and the evaluated platform has since been used for secondary reactions to deposit TiO<sub>2</sub> sols. Thus, TiO<sub>2</sub>-PTFE coatings were made using the simple sol-gel method. Coatings' antibacterial and anti-adhesion efficiency have been measured following exposure tests from Gram-positive and Gram-negative (*Escherichia coli*) bacteria, while their anti-corrosion qualities have been assessed using the electrochemical approach (*Staphylococcus aureus*) bacterial strains. Radiofrequency plasma spray and atmosphere plasma spray were used to apply a PTFE coat to mild steel. According to evaluation results, the subjected scheme produced a friction coefficient that was twice as high under atmospheric plasma treatment conditions as it was when PTFE coating was applied to substrates subjected to low-pressure RF plasma treatment. Steel now has a coat to help prevent wear on exposed appliances in moving machinery. Such parts, such as machine components such as bearings, shafts, gears, and bolts, have improved tribological properties thanks to PTFE coatings. Studies in tribology have been done on either the subsidization of solid lubricant, polymer, or both. According to logical scripture, PTFE surfaces with good wear and corrosion resistance, self-lubrication, low noise, and low-cost relevance are appropriate for use in performance evaluation. Additional schemes to interpret the properties of joints after coat application on balls have been subjected by the decisive consideration of friction and clearance in spherical joints. It has been stated that the kinematic accuracy of mechanisms with joints is important for interpreting joint clearances. The goal is to evaluate how the regional PTFE arrangement of atomic orders by carbon chain distributions and orientations would be affected by thermal differentials. PTFE coat's effects on various substrates. As surgical implants for mechanically designed jointed forms, bases have been proposed. The goal was to understand the development of such coat applicability.

The only recycled material used in the production of virgin PTFE is pure PTFE resin. Virgin PTFE is very chemical resistant, an outstanding electrical insulator, and it maintains its elasticity at low temperatures. Low cost, inability to produce massive sizes, inability to be cemented, ability to change shape under pressure, inability to weld, propensity for creep, and unacceptability under continuous high load are some characteristics of PTFE polymer. Biocompatible PTFE has been issued as a coat additive to TiO<sub>2</sub> (anatase) for surgical implants made of sterile stainless steel.

Molecular dynamics simulations of PTFE helix reversals seen in X-ray diffraction modes. Long-range order is best preserved by large rotations of the helix of the carbon and fluorine chains about the molecular axes. Within a PTFE crystal, helix reversals on nearby chains exhibit inclusive or interacted segregation through clearly defined planes. Due to flaws in the crystal, fluoropolymers now behave differently in terms of phase and mechanics based on chain and segmental motions. The helix act links fluoropolymer mechanical characteristics, phase transitions, rotational disorder, and helix reversals. To improve corrosion resistance in HCL solution, a thin film coating of PTFE has been investigated for deposition on 316L stainless steel. Investigating the morphology of the surface created by the PTFE coating to resist corrosion and reduce exposed current densification. Scheming PTFE coat on machine components made of mild steel as well as on joint balls used in mechanical power transition has provided a replacement for solid lubrication.

Large local rotations about molecular axes have been made possible by PTFE helix reversals that were seen under X-ray diffraction modes without affecting long-range order. Thus, within a PTFE crystal, helix reversals on nearby chains have interacted and segregated onto clearly defined planes. Due to the movement of these planes of defects through the crystal, fluoropolymers' phase behavior and mechanical behavior now depend on on-chain and segmental motions. The impacts of helix reversal activity have been conceptualized in terms of the interplay of helix reversals, rotational disorder, phase transitions, and mechanical characteristics in fluoropolymers. Molecular dynamics simulations of molecular processes have provided insight into the nature of intra- and intermolecular interactions between chain rotations and helix reversals. Low friction PTFE has been useful, but the ball's PTFE covering has increased joint clearance. As a result, more interpretations of the relationship between friction and clearance in spherical joints—that is, the kinematic accuracy of joint mechanisms—have been considered.

## HELIX REVERSAL

Helix reversal activity in polytetrafluoroethylene (PTFE) as defined by molecular mechanics and dynamics simulations shows that helix reversals do form and migrate in PTFE crystals [1]. Helix reversal band formation, which denotes two helix reversals brackets around a small chain segment to oppose helical sense from the parent molecule, has been a defect structure that has developed at low and intermediate temperatures. Half of the helical repeat unit has been the size of the reversal band defect in low and intermediate temperature states. High-temperature states have lessened intermolecular effects. The imposed impetus of helix reversal activity resulted in the rotational disorder. Apart from rotational disorder, it has also been investigated how a chain segment that is surrounded by two helix reversal bands can reorient itself about the molecular axis in the same helical sense as the host chain. The rich phase behavior of polytetrafluoroethylene (PTFE), which is made up of an organic helical chain structure, was created from a chemically simple polymer. PTFE exhibited three solid-state phases, with phase II occurring below 198°C, when operating at atmospheric pressure and within a few tens of degrees Celsius of difference. Upon observation using X-ray and electron diffraction data, this ordered triclinic unit cell contained two chain stems from the opposing hand. The subjective at 198°C has been cited as an order-disorder transition point, and PTFE molecules have a 54/25 u/t (units per turn) helix as opposed to the usual corresponded as a 13/6 u/t ratio. While untwist molecules have adapted to adopt a 15/7 u/t ratio, rotations of molecules about their axes, which have not resulted in helical axes being reoriented, are what cause the disorder. According to X-ray data, the intermediate phase (IV), which was a hexagonal unit cell with a single stem, persisted up to 308°C. Larger cells may have also contained both left- and right-handed helices, according to consistent data from diffraction that was released. Phase I, or the 15/7 u/t ratio helical conformation, has degenerated into a 2/1 u/t ratio planar zigzag form as a result of temperature increases above 308°C, rotational disorder, and untwist helices having suggestive problems. The unit cell maintained its metric hexagonal shape when PTFE was at or below 198°C, and the X-ray results indicated a high degree of order. However, studies using Raman scattering and solid-state Nuclear Magnetic Resonance (NMR) have unequivocally indicated the existence of dynamic disorder in crystalline areas at cryogenic temperatures. Helix reversals have been suggested as

an alternative explanation for the observed presence and mobility of chain faults. PTFE helix reversals have enabled large local rotations about molecule axes without compromising the long-range order observed in X-ray diffraction experiments.

### **NANO COAT ON STAINLESS STEEL**

The antibacterial and anti-corrosion qualities of advanced titanium dioxide-polytetrafluorethylene (TiO<sub>2</sub>-PTFE) nanocomposite coatings on stainless steel surfaces were reported [2]. The use of biomedical metallic implants has frequently failed due to bacterial infection and corrosion. Applying a sol-gel simple two-step dip coat of TiO<sub>2</sub>-PTFE nanocomposite on the stainless-steel substrate has enabled success against bacteria and corrosion. Using a methodical application, a bio-inspired Polydopamine (PDA) sub-layer was formed on a stainless-steel substrate. The resulting TiO<sub>2</sub>-PTFE coat, which was evenly co-deposited onto the PDA sub-layer, had better adhesion and reactivity as a result. Considerable impact on the surface energy of the TiO<sub>2</sub>-PTFE coating has been investigated based on the components of both PTFE and TiO<sub>2</sub>. A matching coat has been obtained with a total surface energy of 26 mJ/m<sup>2</sup>, showing little bacterial adherence to both Gram-positive *Staphylococcus aureus* F1557 and Gram-negative *Escherichia coli* WT F1693. TiO<sub>2</sub>-PTFE coat has demonstrated improved corrosion resistance in bodily fluids when compared to either TiO<sub>2</sub> coat or PTFE coat as a result of the synergisms between the two materials. It has been suggested that the TiO<sub>2</sub>-PTFE coating is biocompatible with fibroblast cells in culture, which supports the use of metallic implants. Despite the fact that deficiencies have been studied as bacteria adhering to implant surfaces, bacteria now become embedded in a dense extracellular matrix to ward off host defence mechanisms like antibiotic penetration.

### **CORROSION INHIBITOR**

Stainless steel (SS) are used frequently in a variety of engineering applications due to their excellent corrosion resistance in various media and fine mechanical properties [3]. Because 316L stainless steel requires a hard coat to withstand wear and corrosion, it exhibits poor corrosion behavior in HCL. Polytetrafluoroethylene (PTFE) has recommended using the spin coating method to coat the 316L SS. Following that, a 40% HCL electrochemical corrosion test was used to modify corrosion resistance in acidic media containing HCL, and potentiodynamic polarization curves were created at room temperature for comparative analysis. While compositional analysis was carried out using energy dispersive X-ray spectroscopy (EDX), scanning electron microscopy (SEM) was used to investigate morphological improvements from specimens before and after corrosion tests. Since the polymer has passivation to current flow, the thickness of the coating reduces the corrosion current density in HCL media, thereby improving the corrosion resistance of the PTFE coating.

### **TRIBOLOGY AFTER COAT**

The surface of heat-treated AISI 1050 carbon steel has been coated with polytetrafluoroethylene (PTFE) polymer material using atmospheric and Radio-Frequency (RF) low-pressure plasma systems. [4] Respected various plasma parameters include the length of the treatment, its frequency, its intensity, and its pressure. A tribological tester has been used to conduct wear tests on coated surfaces. SEM and Optical Microscopy (OM) have both been used to analyze the surface morphology of specimens that have been deposited. According to evaluation results, the subjected scheme produced a friction coefficient that was twice as high under atmospheric plasma treatment conditions as it was when PTFE coating was applied to substrates subjected to low-pressure RF plasma treatment. Steel is frequently used to make engineering tools, such as machine parts like bearings, shafts, gears, and bolts. Because these parts are subject to tribological effects, wear is now secured and prevented by a decrease in coefficient of friction. Techniques for coating implements with PTFE have been designed to distinguish between atmospheric and vacuum plasma classes. While atmospheric plasma uses the atmospheric conditions, vacuum plasma spray systems use an evacuated chamber. Plasma discharges similar to those used to deposit PTFE on Al or Cu surfaces under atmospheric conditions were used in the methods for coating carbon steel with the material. Atmospheric pressure cold plasma jets can be generated by small

radiofrequency plasma jet sources. Treatment with atmospheric plasma has improved adhesion and increased hydrophilicity.

### COAT ON SPHERICAL JOINT

The frictional resistance and practical joint clearance have both been reduced by the application of Polytetrafluoroethylene (PTFE) coatings to the surface of spherical joints [5]. Investigating different solid lubricants has mandated through an initial experiment about assessment of friction performance by as such lubricants, then selective solid assess lubricity has scoped to estimate as friction-reducing layer fill within the gap between the ball and the support in a joint system. Since solid lubricant coated spherical joints have undergone periodic rotating abrasion tests, PTFE coating has successfully decreased ball joint clearance, shielded component surfaces from quick wear, and reduced friction [6-10]. The mechanism's reduced spherical joint clearance and increased friction resistance as a result of its functional performance make adding an addition to the design impossible. Additional schemes to interpret joint properties have been subjected by the decisive consideration of friction and clearance in spherical joints. It has been stated that the kinematic accuracy of mechanisms with joints is important for joint clearances. An established contact model of joints with clearance has been supplemented with a four-bar linkage model that analyzes linkage kinematic performance under different clearances.

### CONCLUSION

This investigation confirms PTFE's multifaceted potential as an engineering polymer with valuable tribological and protective properties. PTFE's molecular structure, marked by helix reversals and phase transitions, underpins its stability and adaptability across a range of temperatures and applications. The integration of PTFE with TiO<sub>2</sub> in nanocomposite coatings offers enhanced antibacterial and anticorrosive properties, providing substantial benefits in medical implant technology. Plasma coating techniques further enhance PTFE's performance in high-wear environments, positioning it as an ideal candidate for coating machinery components and medical devices. These insights into PTFE's phase behavior and coating applications pave the way for future developments in materials engineering, expanding the scope of PTFE in high-performance, high-durability applications. Further research into optimizing PTFE's tribological and thermal characteristics will be instrumental in expanding its industrial and biomedical applications.

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