

Advanced Coating Techniques: Improving Metal Longevity through Nanocomposite Applications

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Abstract

In order to protect metal and lessen wear and tear, this topic has scriptive issues pertaining to coatings that are applied to metal surfaces. Components in slurry pumps have better wear and abrasion resistance compared to their surface properties. An interface region has been used to separate at least two immiscible phases in the material that makes up a nanocomposite coating. Fillers have become a major constituent of the matrix, wherein both ingredients have arranged from dimension up to the nanoscale in any one of the axes. Depending on different factors, such as (a) the kind of nanostructured filler and (b) the type of nano-matrix where the nano-fillers have dispersed, different categories of nanocomposite coatings have different rated clips. The nanocomposite coats have been categorized into three groups according to the type of nanostructured filler used: 0D, which uses nano-3D particles as filler; 1D, which uses nano-2D tubes or whiskers as filler; and 2D, which uses nano-1D layers as filler. Matrix/nanofiller nanocomposite coat groups, for example, are denoted by O/I, O/O, I/O, and I/I. Similarly, groups of nanocomposite coat based on matrices have conclusive from either organic (O) or inorganic (I) combinations. The developed surfaces that were scoped in the case of the ash handling provisions, where erosive wear has also translated to chemical based on composition differential, have been returned to ethical evaluation. One-step and two-step laser depositions have been produced through research on thick laser coatings.

Keywords: Nanocomposite coatings, metal protection, wear resistance, corrosion resistance, thermal stability

INTRODUCTION

The process of one-step laser deposition (1SLD) involves injecting powder into a laser beam, heating it to the point of melting, and then projecting the melted laser plum onto the substrate. This creates a molten interfacial that results in instanced reticulation. Plasma transferred arc (PTA) welding is the name given to the methodically differentiated process that suggests a synergistic effect of the welding deposition technique. Rapid prototyping by 1SLD is an example of an industrial application; thus far, one step processing has been successful in producing solid pieces with intricate 3D shapes. Following the formation of fine-grain and dispersion strengthening, the nano-indentation hardness and corrosion

resistance of the Stellite 6 coat added with La₂O₃ have improved. It has been possible to prepare Ni-based WC coating on 4145H steel using the PTA method with different weight percentages of Y₂O₃ (0 to 1%). Success has been determined by verifying the impact of Y₂O₃ addition on phase composition, mechanical properties, and microstructure. Materials now have a range of protections added to them to stop material components from corroding. Matrix materials are categorically derived from either organic or inorganic matter. As is customary, organic chemical companies have provided polymer-based nanocomposite matrices made of

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epoxy, polyurethane, Chitosan, polyethylene glycol (PEG), polyvinylidene fluoride (PVDF), polyaniline (PANI), polypyrrole (PPy), polystyrene, polyamic acid, and polyimide, as well as rubber-modified polybenzoxazine (PBZ). These polymers include reactive trimethoxysilyl (TMOS), pullulan, fluoro acrylic polymer, ethylene tetrafluoroethylene (ETFE), polyacrylate, poly (N-vinyl carbazole), polycarbonate, fluorinated polysiloxane, polyester, polyacrylic, polyvinyl alcohol (PVA), polydimethylsiloxane, polyamide, and UV-curable polymers.

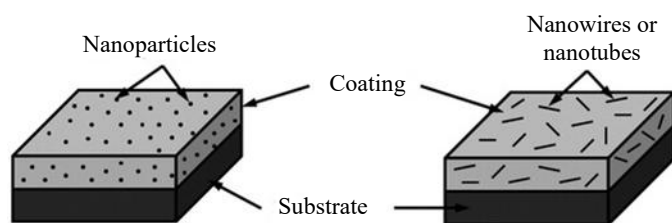
The success of metal/alloy matrix-based nanocomposite coatings has been surpassed by inorganic matrix. Chemical vapour deposition (CVD), powder metallurgy, thermal plasma spray, sol-gel, epitaxial growth, cold spray, and electrodeposition are just a few of the techniques that have been used. In metal matrix composite coatings, the dispersed second phase has been linked to wear, magnetic properties, and resistance to oxidation and corrosion. Nanoparticles, such as carbides, nitrides, borides, oxides, metallic particles, clay, carbon nanotubes (CNT), and nanodiamonds, have been used as materials for inorganic nano-fillers inclusive into metallic matrices. Figure 1 Other than that, PTFE, PEO, PANi, nano-cellulose, and cellulose nanocrystal have been used as organic nanoparticle fillers. Thus, various processes, such as spraying, sputtering, and electroplating or electroless plating, have been successful in applying coats made of suit-an-addendum materials.

Coating to Scope Functional Prospect

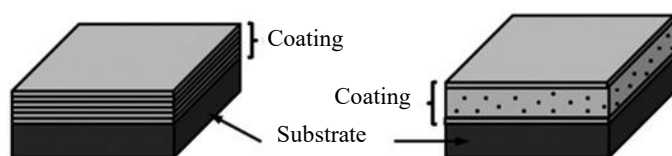
Coatings are used to provide combined frames on surfaces for decorative, protective, and functional purposes. Decision-making composition must help to address deficiencies in the actual application of coated substrate; for example, classical properties extended to functional implication adopted as self-cleaning, easily cleaned, i.e., antibacterial, anti-graffiti, and soft feel. The functional coatings that are used have the usual applicability versatility and the potential to link with other requirements. For example, non-stick cookware coatings need to be resistant to heat, abrasion, and scratches. Extended consideration was given to the capabilities that Typified used to be scheduled, including cost effectiveness, easy application, durability, reproducibility, tailored surface morphology, and scripture echo friendliness. Otherwise, the basis for a categorically studied fame is function-specific characteristics. Their mechanical, chemical, thermal, and physical properties are average.

Anticorrosive Coatings

When exposed to the elements, rust-prone iron or steel have been coated with anticorrosive materials. Rust is sometimes referred to as corrosion, and corrosion has been connected to harmful material-environment interactions. Corrosion is a harsh term typically used to describe reactions to metals, but environmental degradation can also apply to nonmetallic substrates like plastics, concrete, wood, etc.



(a) Matrix-reinforced nanocomposite coatings



(b) Layered nanocomposite coatings

Figure 1. Nanocomposite Coatings.

Thermal and Fire-Resistant Coating

For metallic substrates used in boilers, grills, and nonstick cookware, high heat-resistant coatings have been made available. Products containing silicon or fluorine have been known to exhibit high thermal resistance. Fluorinated coatings have been found to be suitable for use up to approximately 300 °C, beyond which they have been found to produce harmful byproducts. However, silicon containment in coatings has been more successfully demonstrated because silicon-containing polymers have been better in thermal resistance linked to suggestive cleavable margins, i.e., cleavage resistance of silicon in accordance of bonds compared to that of carbon bonds in analogous molecules. Binder originated differential produces from phenolic or epoxy has been studied instead. Alternatively, advancements related to silicon-based coatings, such as siloxanes or inorganic silicates, have been assigned to confer thermal resistance up to 1000°C. Because of their state of asset as to set being associations from blends of silicones with acrylate, epoxy, or urethanes, organic and inorganic silicon have been expensive.

Scratch and Abrasion Resistant Coating

The ability of coated materials to maintain their aesthetic appearance has been tested with scratch and abrasion resistant coatings, which demonstrate resistance to damage caused by scratches and/or abrasions. Cars with clear aesthetic coats have good resistance to scratches and abrasions, which is in line with the ability of the structures to withstand damage that would otherwise cause damage to the underlying substrate. In any other case, the policy has limited under margins in order to enhance scratch resistance without compromising other qualities. According to logical apt development, adding more accessible cross links to the coat binder improves scratch resistance. However, increasing cross links, or the hardness of the films, decreases impact resistance and the effectible margin of flexibility.

Self-Healing Coating

The self-cleaning characteristic of lotus leaves, which results from their unique surface morphology and hydrophobicity, has inspired the development of self-cleaning coats. This useful characteristic, known as the "Lotus Effect" or "self-cleaning," refers to the way that surfaces exposed to soil have taken on the ability to clean themselves through precipitation.

Antibacterial Coating

Antibacterial coatings have demonstrated resistance against microorganisms, including bacteria, fungi, viruses, and normal decay-related hazards to personal hygiene. Microbes frequently grow on coats under exposure, which can lead to aesthetic issues like coating discoloration or, in the case of metal substrates, surface damage from biofilm formation, also known as microbe corrosion. Biocides-containing microcapsules have been a developed solution, whereby the longevity and efficacy of antimicrobial coatings have been improved. Inorganic biocides include silver, zinc oxide (ZnO), copper oxide (CuO), TiO₂, and selenium; biocides have included polymers, tertiary alkyl amines, and organic acids.

Antifouling Coating

Antifouling coatings are designed to withstand marine organisms, which pose a threat to any objects used in a marine environment. The term "foul" originated from the unwanted growth or deposition of these organisms on objects. Coastal waters have seen a greater prevalence of fouling than deep sea exposure. Otherwise, fouling has been more of an issue for ships and boats that are docked or move slowly than for those that are exposed to deep water. Failure has been divided into two categories by types of marine organisms involved in fouling: micro fouling and macro fouling, wherein marine animals like barnacles and tubeworms and plants like algae have been subscribed to.

Nano Polymer Coating

A variety of composite films have employed nanopolymer coatings as their host matrices. Described organic or inorganic particles can be mixed with or incorporated into conducting polymers, altering their appropriate morphology, conductivity, and physical properties based on the appliance's need for corrosion protection. Thus, the concept of mixed ionic/electronic conductivity originated because conducting polymers have been categorized according to inclusive electrochemical properties.

Self-Assembled Nanophase Coating

Traditional sol-gel method and hydrolysis-oriented condensation processes have been used in self-assembled nanophase coatings, with condensation polymerization coming next to address film applicability. The development of voids and channels throughout solid gel, where crack-formation has not reticulated sufficient corrosion protection, has limited the evaporation process. Instead, a maximum of 2 mm has been established as the maximum coat thickness that can be achieved due to limitations in sol-gel technology.

Self- Cleaning and Biocidal Coating

Research has been done on self-cleaning and biocidal coatings to offer both biocidal activity and easy cleaning; otherwise, self-clean composition. The ability of the coat to be biocidal and self-clean has been achieved through the addition of suit-an-addendum nanoparticles, such as titanium oxide and silver. Paints that are antibacterial, anti-ultraviolet, and self-cleaning have all been created using nano TiO₂. The term "self-clean act" refers to a material's ability to withstand the inclusion of hydrophobic properties, which allow water droplets to bead off and pick up dirt and contaminants as they move.

Characteristics of Coat to Favour

Nguyen-Tri Phuong et al. [1] clarified the characterization of coatings for nanocomposite materials. After adding nano-fillers to organic coatings, research has shown that there is a decrease in porosity and zigzag diffusion path for harmful species. The addition of nano-filler to corrosion-resistant coat matter has increased barrier properties and decreased delamination or coat blistering. After adding rated inclining/declining containment of hard nanocrystalline phases within a metallic matrix, additive advent has gradually increased/decreased the hardness of metal deposited coatings. Application has been changed to include intrinsically sensitive antifouling/antibacterial, superhydrophobic, anticorrosion, self-cleaning, and inexpensive surfacing specifically for the electronics industries.

Prevention of Slurry Erosion

The effects of tribological parameters on the slurry erosion behavior of coated and uncoated materials were clarified by Sunil Kumar and Jasbir Singh Ratol [2] The ash slurry disposal system of thermal power plants has been adopted in studies of the slurry erosion behavior of coated and uncoated materials in various industrial applications. Wear has been most noticeable on important parts like the impeller and housing of the ash slurry disposal pumps in thermal power plants. Rated slurry erosion has been assessed through the measurement of tribological parameters, including solid particle concentration, impact velocity resulting from eroded target surface, impact angle, eroded particle size and shape, hardness of striking particles and target material, etc. Tribological factors have an impact on the significance of slurry erosion; in the absence of these factors, slurry resistance has been methodically achieved following a thorough assessment of the material or coat. As a result, slurry erosion has been reduced through careful material selection and/or the application of an appropriate protective coating, which improves the surface qualities of the material. Slurry disposal pumps, which are made up of an impeller and a volute casing in thermal power plants, are responsible for a portion of the deficit power generation associated with slurry erosion. The runner, guide vanes, needles, and seats of hydraulic turbines in hydroelectric power plants have also seen the effects of erosion. The suggested scope has been to protect the system for maximum utility by either complete prevention or early detection.

In thermal power plants, combined slurry is made by combining bed ash, fly ash, and clean water in a mixing tank. Once the desired combined slurry density is reached, the slurry is centrifugally pumped out of the plant. Target materials that have been subjected to production include mild steel, brass, nickel-hardened alloy steels, chromium-alloyed cast iron, stainless steel, and aluminum alloy. However, erodent solid particles have been used in industrial settings for a variety of purposes. These include clays, volcanic ash, fly ash, bed ash, silica sand, quartz, feldspar, muscovite, garnet, and silicon carbide. The variation up to mark from ash to surface property correspondences has been prompted by an aloft that may or may not have raised ash content. This has minimized wear depending on surface properties. Table 1 displays the minerals and trace elements found in the ash of Indian coal.

Table 1. Compose to form Indian coal ash. [2]

S. N.	Coal	Ash content (wt. %)	Major minerals	Trace minerals
1.	Indian coal A	46.7	Quartz, Illitic clay, Raolinite, Muscovite, Siderite	Barites, Rutile, Feldspar, pyrite, Ilmenite, Zircon
2.	Indian coal B	30.7	Quartz, Illitic clay, Feldspar, Kaolinite, Muscovite, Siderite	Apatite, Garnet, Ilmanite, Rutile, Monazite, Zircon
3.	Indian coal C	45.6	Quartz, Illitic clay, Feldspar, Garnet, Muscovite, Kaolinite	Apatite, Rutile, Ilmenite, Pyrite, Monazite, Zircon

The slurry pumps are set up in series to enable the hydraulic disposal of a massive volume of ash from the power plant to the ash pond. The impeller and casing of ash handling slurry pumps should be made of highly wear-resistant materials, such as nickel-hardened alloy steels, chromium-alloyed cast iron, etc. When using low-strength materials, properties can be improved by coating the intended surface with a material that resists wear. In this context, several coating techniques are compared and discussed.

REDUCE CORROSION ON STEEL FASTENER

The properties of coatings and their deposition processes on steel fasteners were reviewed by Ping CHUNG et al. [3] Metals have been joined by mechanical fasteners in the building, construction, automotive, and aerospace industries. The subject has evaluated how mechanical fasteners corrode and offered strategies for extending the service life of steel fasteners by applying protective coatings to these parts. Proposals to connect such abrasion from coat scaled simplex have been adopted with a standard scope, such as mechanical plating, ion vapour deposition, hot-dip galvanizing, and electroplating. As a result, composition research has been used to evaluate suit-in-scale composed formation. Thus far, the addition of microstructural characteristics such as morphology and porosity, resistance to corrosion, and friction-causing substances has confirmed the effectiveness of coatings. The concept of selective coatings has also been based on hydrogen embrittlement, which has been another unexpected problem regarding the early failure of high strength steel fasteners. According to ratings given to both stationary and mobile infrastructures, sustainable fasteners are responsible for enhancing the long-term mechanical or structural integrity of assembled parts and systems. Alternatively, coated fasteners allow for a service life of at least 30 to 50 years in the building and construction sector and up to 20 years in the automobile sector. In order to obtain scriptive joining techniques assigned, lightweight design and production in automotive structures have been the strategy, followed by coated fasteners and coating materials. High strength alloys of magnesium and aluminum, ultra-strong steels, and hybrid composites have all been studied for schemed compositions. In service exposure, atmospheric corrosion, galvanic corrosion, crevice corrosion, and embrittlement have all been described as descriptive issues with steel fastener degradation. Galvanic corrosion has caused coating perversion because of the vast disparity in the electrochemical potentials of the materials in their fastened states. If not, the electrolyte or conducting medium contains high-strength steel fastener assembly and aluminum alloy sheeting, among other scripts. The most common kind of corrosion has been localized attack on metal surfaces, such as the crevice between two joining surfaces, which is known as crevice corrosion. The small space between the metal sheets and fasteners has been a citation-specific problem. There have been suggested differences in electronegativity between the electrolyte concentration outside the crevice (high oxygen content) and inside the crevice (low oxygen content). High strength steel fasteners typically fail under tension due to hydrogen embrittlement, which suggests that the stress value applied during failure was significantly lower than the steel's specified yield strength.

Laser Applicability to Coat by One or More Steps

Thick laser coatings were explained by L. Pawlowski [4] through the use of laser beams. The interaction between laser beams and typical coating materials has been described for coat deposition by laser plume processes in the conditional atmosphere. One-step or two-step deposition of coats has been provisionally represented by characteristic laser light beams. In addition to replacing thermal spray, lasers have been used as diagnostic and technical tools. As a result, process control has benefited from

the use of laser Doppler velocimetry (LDV) or laser two-focus systems (L2F) to determine the velocity of sprayed particles. Lasers are intended to create high-quality spray deposits, or coatings, and are intended to evaluate the quality using non-destructive testing methods (NDT). In the context of thin film processing, lasers have been used in conjunction with thermal spraying to create chemical vapour deposition (CVD), physical vapour deposition (PVD), pulsed laser deposition (PLD), and other effects.

12. 11. 6. Coat Compose to Adhere Presence of Rare Earth Element

The effects of Y₂O₃ on the microstructure and wear resistance of WC/Ni composite coatings made using plasma transferred arc were clarified by Yue Hou et al. [5]. Due to their exceptional strength, superior corrosion resistance, and high temperature oxidation resistance, pure nickel and its alloys are now widely used in industrial and marine processing. However, under severe friction conditions, performance is limited by lower hardness and poor tribological performances. Ni-based coatings containing carbide ceramic powder, such as WC and TiC, have improved wear resistance; as a result, after surface modification, definite employment has become possible. It has long been standard practice to modify a metal or alloy's surface using chemical vapour deposition, thermal spraying, plasma transferred arc (PTA), and laser cladding techniques to prevent corrosion and increase wear resistance. The PTA technique has been methodical in producing high-performance composite coatings because it was adopted due to its wide applicability, low equipment cost, ease of use, and high energy density. A deficiency has arisen due to the formation of cracks in the Ni-based ceramic composite coat exposed to service. Studies have been conducted to reduce the effects of Y₂O₃'s cracking sensitivity on the microstructure and wear resistance of WC/Ni composite coatings. The ability to refine grain size through inclusive opportunity from rare earth has optimized the microstructure of ceramic bonding. The surface of 4145H steel now has WC-reinforced nickel composite coatings made possible by the application of the plasma transferred arc (PTA) technique. By using scanning electron microscopy (SEM) with an energy-dispersive spectrometer (EDS) attachment, X-ray diffraction (XRD), microhardness testing, and wear experiments, researchers have examined the effects of varying Y₂O₃ content on microscopic morphology, phase composition, micro-hardness, and wear behavior. From the time coatings made of WC-reinforced Ni59 composite coatings were deposited until different composition constitutive succeeded following the addition of 0, 0.5, and 1 weight percent Y₂O₃, subjectivity has been inclusive. Coarse dendrites and the microstructure's grain refinement have improved following inclusive in phase composition, which is provided by the hard phases of WC and W₂C, the Ni, FeNi₃ solid solution, and the (Fe, Ni) eutectic phase. Due to the uniform organizational structure in comparison to the substrate and other coatings, there have been logical increases in wear resistance as a result of the respected micro-hardness improving from 500 HV to 800 HV. A significant decrease in the coefficient of friction due to wear has been used to assess the improvement of the rare earth Y₂O₃ addition.

Functionally Hard Coat

Alicja Kazek-Kęsik and Vlad Diciuc [6] clarified the process of structural inspection following the development of functional hard coatings. Depending on their intended use, turning tools come in a variety of sizes, shapes, bases, coatings, and geometric configurations. The usual use of turning tools has been in machining. The reaction of coated high-speed steel (HSS) turning tools to hard coatings that develop on the tool material has been investigated. The tool's functional scope has been methodically developed, with functional grades based on the tool's ultimate application. The development of a hard coat layer on actual cutting tools has been reported, in place of a massive global expenditure increase of 16.33 billion US dollars. According to statistical estimations, 39% of industry cutting tools are milling tools and 30% are turning tools. The annual growth rate of this industry has increased by 4-5%. According to statistics, appliance-specific issues have been 53% cemented carbide and 20% high-speed steel (HSS). According to statistics from 2000, coatings on cutting tools have an applicability of 50% on HSS tools, 85% on carbide tools, and 40% on extremely hard tools. HSS turning tool coatings offer protection against adhesion, oxidation, and abrasion. They also introduce the use of a heat barrier to facilitate heat flow from the cutting area to the tool matter. In contrast to continuous cutting tools, heat

insulation techniques have increased productivity and extended the life of tools. The coat structure formation method has been studied, and it has been found that while cut apply has specific advantages and limitations, the same coatings are likely to be applicable to both HSS and carbide tools. As a result, the properties of the tool surfaces have improved correspondingly, leading to increased productivity in the manufacturing processes. The creation of different hard coatings has been linked to ethical turning tool destination origins because of the shape of the turn tool, which adopted coat techniques liable for layer formation by physical or chemical vapour deposition. Chemical vapour deposition (CVD) or physical vapour deposition (PVD), arc ion plating, ion implantation, plasma-enhanced CVD (PECVD), and magnetron sputtering have been the conventional techniques used to apply hard coat. The PVD technique has allowed for the coat formation of various materials, such as Ti, Al, Cr, Ni, N, O, Si, and C, to occur within 1–5 μm . In response to the formation of a 4–20 μm thick coat of narrow range materials (Ti(C,N), TiN, and Al_2O_3), suggestive CVD has been studied based on control parameters such as stress state, phase composition, and crystal orientation. Based on HV, 0.05 tests, the justified adjusted coat is controlled under the hardness margin between 2300 and 5500. The dependence of the coat's mechanical properties, oxidation resistance, and thermal resistance on its structure, phase composition, and homogeneity has been studied. The limits of charge from PVD and CVD have historically been high, peaking at 950°C for layer deposition and low layer adhesion to substrate. As a result, interface engineering has firmly established scriptural interlayer formation techniques, such as functionally graded formation. In order to prevent irregularities, guidelines have been applied to coats regardless of any deviations in the destination or surface roughness before or after coat application.

Modification by Phosphatising

The modified manganese phosphate conversion coating on low-carbon steel was clarified by Jakub Duszczyk et al. [7] The chemical reactions between the metal surface and the phosphate liquor environment cause scriptive problems for primary galvanic conversion coatings, which are used to prevent corrosion in steel structures. To improve a particular quality of the finished coating, new metal cations have been added to the phosphate reaction environment through secondary phosphating. Application of the Zn surface led to the introduction of phosphate coats, which have unique problems as a result of the exploration of manganese and iron, also known as manganese coats. Regular examination using an electron microscope (SEM), X-ray diffraction (XRD), and electrochemical tests has been associated with structural conformations. The process of creating mn-coat involves reacting modified phosphate baths with different metals, such as Ba, Zn, Cd, Mo, Cu, Ce, Sr, and Ca tools. Manganese phosphate crystals have been identified as a suggestive structure; however, the destructive effects of chromium and molybdenum have been limited.

Multi-Layered Coat from Investigative Aptitude

The multi-layered coat for high-temperature steam oxidation was clarified by T. Dudziak et al. [8] using thermogravimetric analysis up to 1000 oC. The performance of steam oxidation has been attributed to multi-layered coatings applied to 800H Incoloy and 310 stainless steel. The applicability of the aluminized thermal diffusion coat technique has been exchanged, meaning that the option to form aluminides instead of a thermal barrier monolithic coat is now available. Chronology has thus been used to accomplish more top surfacing of particular compositions. Thermogravimetric analysis equipment has been used to examine suggestions regarding various coating systems in relation to uncoated metals. Test specimens were exposed to corrosive temperatures of 1000 oC during the steam oxidation stage. It has been suggested that coat systems can withstand high-temperature steam oxidation without experiencing breakaway degradation. The determination of sample mass change, microanalyses with light optical microscopy, scanning electron microscopy (SEM) fitted with energy x-ray dispersive spectrometry (EDS), and microhardness tests have all been prompted by experiment schedules.

Dip-Coat to Deposit Fibrous Matter

For fibrous materials, Xiaoning Tang and Xiong Yan [9] clarified the dip-coat technique. The functional scope of the dip-coat method of film deposition is applicable to fibrous matter, where

different process parameters are based on thickness. The possibilities of modified dip-coating methods to create deposited films for fibrous substrates have been expanded upon through the use of convected and sol-gel based techniques. The developed front from methodical vacuum-assisted, spin-assisted, photo-assisted, and multi-layered dip-coating methods has also been highlighted by the cited schemes. Coated fibrous materials can be used in a variety of applications, such as conductive textiles, fibrous-based energy storage devices, photonic crystals, self-cleaning textiles, etc., exposed medium, and so on. Therefore, the initial goal of dip-coat applicability has been to enhance wave-ability, spin-ability, abrasion resistance, and mechanical properties.

Nano-Metric Organic-Coat

Nanotechnology was clarified by A. Mathiazhagan and Rani Joseph [10] to write a new viewpoint in organic coat technique. Artificially created particle tubes or layers with a diameter of less than 100 nanometers (nm) in at least one dimension—length, width, and breadth—have unique characteristics not found in conventional practices. Various coatings, such as paint formulations using different pigments, have applied nanomaterials; otherwise, susceptible sets may encourage the addition of nanomaterials or fillers to the coat.

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